

L-Cysteine, N,S-di(trifluoroacetyl)-, 2,2,3,3,3-pentafluoropropyl ester

Inchi:	InChI=1S/C10H6F11NO4S/c11-7(12,10(19,20)21)2-26-4(23)3(22-5(24)8(13,14)15)1-27-6
InchiKey:	DCQLIBWHNMOBAW-UHFFFAOYSA-N
Formula:	C10H6F11NO4S
SMILES:	O=C(OCC(F)(F)C(F)(F)F)C(CSC(=O)C(F)(F)F)NC(=O)C(F)(F)F
Mol. weight [g/mol]:	445.21

Physical Properties

Property code	Value	Unit	Source
gf	-2469.92	kJ/mol	Joback Method
hf	-2821.84	kJ/mol	Joback Method
hfus	37.57	kJ/mol	Joback Method
hvap	59.20	kJ/mol	Joback Method
log10ws	-3.91		Crippen Method
logp	2.596		Crippen Method
mcvol	208.140	ml/mol	McGowan Method
pc	1755.07	kPa	Joback Method
rinsol	1140.50		NIST Webbook
rinsol	1140.50		NIST Webbook
tb	709.79	K	Joback Method
tc	881.38	K	Joback Method
tf	462.71	K	Joback Method
vc	0.869	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	610.30	J/molxK	709.79	Joback Method
cpg	618.70	J/molxK	738.39	Joback Method
cpg	626.38	J/molxK	766.99	Joback Method
cpg	633.40	J/molxK	795.59	Joback Method
cpg	639.79	J/molxK	824.18	Joback Method
cpg	645.61	J/molxK	852.78	Joback Method
cpg	650.92	J/molxK	881.38	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U352347&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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