

«beta»- Alanine, N-(2,6-difluorobenzoyl)-, tetradecyl ester

Inchi:	InChI=1S/C24H37F2NO3/c1-2-3-4-5-6-7-8-9-10-11-12-13-19-30-22(28)17-18-27-24(29)2
InchiKey:	OOQPBINVCQVNML-UHFFFAOYSA-N
Formula:	C24H37F2NO3
SMILES:	CCCCCCCCCCCCCOC(=O)CCNC(=O)c1c(F)cccc1F
Mol. weight [g/mol]:	425.55

Physical Properties

Property code	Value	Unit	Source
gf	-418.72	kJ/mol	Joback Method
hf	-1021.23	kJ/mol	Joback Method
hfus	66.82	kJ/mol	Joback Method
hvap	93.32	kJ/mol	Joback Method
log10ws	-8.04		Crippen Method
logp	6.329		Crippen Method
mvol	347.790	ml/mol	McGowan Method
pc	983.31	kPa	Joback Method
rinpol	3317.00		NIST Webbook
rinpol	3317.00		NIST Webbook
tb	964.03	K	Joback Method
tc	1181.31	K	Joback Method
tf	587.63	K	Joback Method
vc	1.373	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1163.99	J/mol×K	964.03	Joback Method
cpg	1180.58	J/mol×K	1000.24	Joback Method
cpg	1195.82	J/mol×K	1036.46	Joback Method
cpg	1209.76	J/mol×K	1072.67	Joback Method
cpg	1222.47	J/mol×K	1108.88	Joback Method
cpg	1234.00	J/mol×K	1145.09	Joback Method
cpg	1244.40	J/mol×K	1181.31	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U321851&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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