

Bicyclo[3.3.2]decane

Inchi:	InChI=1S/C10H18/c1-3-9-5-2-6-10(4-1)8-7-9/h9-10H,1-8H2
InchiKey:	WMRPOCDOMSNXCQ-UHFFFAOYSA-N
Formula:	C10H18
SMILES:	C1CC2CCCC(C1)CC2
Mol. weight [g/mol]:	138.25
CAS:	283-50-1

Physical Properties

Property code	Value	Unit	Source
chs	-6343.40 ± 7.10	kJ/mol	NIST Webbook
gf	106.42	kJ/mol	Joback Method
hf	-106.00 ± 7.50	kJ/mol	NIST Webbook
hfs	-164.00 ± 7.10	kJ/mol	NIST Webbook
hfus	9.53	kJ/mol	Joback Method
hsub	58.00	kJ/mol	NIST Webbook
hsub	58.00 ± 2.00	kJ/mol	NIST Webbook
hvap	38.37	kJ/mol	Joback Method
log10ws	-3.31		Crippen Method
logp	3.367		Crippen Method
mcvol	130.040	ml/mol	McGowan Method
pc	3025.61	kPa	Joback Method
tb	458.76	K	Joback Method
tc	680.81	K	Joback Method
tf	224.26	K	Joback Method
vc	0.477	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	401.24	J/mol×K	680.81	Joback Method
cpg	385.09	J/mol×K	643.80	Joback Method
cpg	367.80	J/mol×K	606.79	Joback Method
cpg	349.32	J/mol×K	569.78	Joback Method
cpg	329.59	J/mol×K	532.78	Joback Method

cpg	308.56	J/mol×K	495.77	Joback Method
cpg	286.16	J/mol×K	458.76	Joback Method
dvisc	0.0046703	Paxs	224.26	Joback Method
dvisc	0.0004040	Paxs	458.76	Joback Method
dvisc	0.0005024	Paxs	419.68	Joback Method
dvisc	0.0006533	Paxs	380.59	Joback Method
dvisc	0.0009023	Paxs	341.51	Joback Method
dvisc	0.0013547	Paxs	302.43	Joback Method
dvisc	0.0022946	Paxs	263.34	Joback Method
hsubt	58.00 ± 2.00	kJ/mol	290.00	NIST Webbook

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C283501&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

chs:	Standard solid enthalpy of combustion
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfs:	Solid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hsub:	Enthalpy of sublimation at standard conditions
hsubt:	Enthalpy of sublimation at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point

vc: Critical Volume

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