

5-Undecyne

Other names:	5-C ₁₁ H ₂₀
Inchi:	InChI=1S/C ₁₁ H ₂₀ /c1-3-5-7-9-11-10-8-6-4-2/h3-9H2,1-2H3
InchiKey:	VRQLDBSWBBKOCR-UHFFFAOYSA-N
Formula:	C ₁₁ H ₂₀
SMILES:	CCCCC#CCCCC
Mol. weight [g/mol]:	152.28
CAS:	2294-72-6

Physical Properties

Property code	Value	Unit	Source
gf	244.54	kJ/mol	Joback Method
hf	1.93	kJ/mol	Joback Method
hfus	27.37	kJ/mol	Joback Method
hvap	42.23	kJ/mol	Joback Method
ie	9.11 ± 0.02	eV	NIST Webbook
log10ws	-4.22		Crippen Method
logp	3.760		Crippen Method
mcvol	157.250	ml/mol	McGowan Method
pc	2237.64	kPa	Joback Method
ripol	1104.00		NIST Webbook
ripol	1127.00		NIST Webbook
ripol	1104.00		NIST Webbook
ripol	1110.00		NIST Webbook
ripol	1110.00		NIST Webbook
ripol	1109.00		NIST Webbook
ripol	1104.00		NIST Webbook
ripol	1109.00		NIST Webbook
ripol	1104.00		NIST Webbook
ripol	1287.00		NIST Webbook
ripol	1286.00		NIST Webbook
ripol	1286.00		NIST Webbook
ripol	1286.00		NIST Webbook
ripol	1286.00		NIST Webbook
ripol	1280.40		NIST Webbook
ripol	1266.00		NIST Webbook
ripol	1278.80		NIST Webbook
ripol	1263.60		NIST Webbook

ripol	1277.50		NIST Webbook
ripol	1261.10		NIST Webbook
ripol	1286.00		NIST Webbook
tb	471.20	K	NIST Webbook
tb	469.00 ± 2.00	K	NIST Webbook
tb	469.00 ± 2.00	K	NIST Webbook
tb	471.00	K	NIST Webbook
tc	642.86	K	Joback Method
tf	319.83	K	Joback Method
vc	0.614	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	331.56	J/mol×K	460.08	Joback Method
cpg	347.02	J/mol×K	490.54	Joback Method
cpg	361.85	J/mol×K	521.01	Joback Method
cpg	376.06	J/mol×K	551.47	Joback Method
cpg	389.69	J/mol×K	581.93	Joback Method
cpg	402.74	J/mol×K	612.39	Joback Method
cpg	415.23	J/mol×K	642.86	Joback Method

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	351.00	K	1.30	NIST Webbook
tbrp	351.50	K	1.30	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54154e+01
Coeff. B	-4.30826e+03

Coeff. C	-7.21300e+01
Temperature range (K), min.	356.92
Temperature range (K), max.	498.52

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C2294726&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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