

# Benzoic acid, 3,5-dinitro, 2-methylpropyl ester

<b>Inchi:</b>	InChI=1S/C11H12N2O6/c1-7(2)6-19-11(14)8-3-9(12(15)16)5-10(4-8)13(17)18/h3-5,7H,6
<b>InchiKey:</b>	BIDNGDDDZYVORY-UHFFFAOYSA-N
<b>Formula:</b>	C11H12N2O6
<b>SMILES:</b>	CC(C)COC(=O)c1cc([N+](=O)[O-])cc([N+](=O)[O-])c1
<b>Mol. weight [g/mol]:</b>	268.22
<b>CAS:</b>	10478-01-0

## Physical Properties

Property code	Value	Unit	Source
gf	-30.37	kJ/mol	Joback Method
hf	-328.38	kJ/mol	Joback Method
hfus	39.49	kJ/mol	Joback Method
hvap	85.63	kJ/mol	Joback Method
log10ws	-4.03		Crippen Method
logp	2.316		Crippen Method
mcvol	184.370	ml/mol	McGowan Method
pc	2784.72	kPa	Joback Method
rinpol	1897.00		NIST Webbook
rinpol	1910.00		NIST Webbook
rinpol	1983.00		NIST Webbook
rinpol	1903.00		NIST Webbook
rinpol	1889.00		NIST Webbook
rinpol	1903.00		NIST Webbook
ripol	2804.00		NIST Webbook
ripol	2825.00		NIST Webbook
ripol	2805.00		NIST Webbook
ripol	2788.00		NIST Webbook
tb	867.25	K	Joback Method
tc	1121.93	K	Joback Method
tf	609.57	K	Joback Method
vc	0.726	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	534.59	J/mol×K	867.25	Joback Method
cpg	544.76	J/mol×K	909.70	Joback Method
cpg	553.82	J/mol×K	952.14	Joback Method
cpg	561.80	J/mol×K	994.59	Joback Method
cpg	568.73	J/mol×K	1037.04	Joback Method
cpg	574.64	J/mol×K	1079.48	Joback Method
cpg	579.57	J/mol×K	1121.93	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10478010&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10478010&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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