

# Androst-4-ene-3,17-dione, 11-hydroxy-, (11 «beta»)-

<b>Other names:</b>	Androst-4-ene-3,17-dione, 11 «beta»-hydroxy- u 2826 11 «beta»-Hydroxy-4-androsten-3,17-dione 4-Androstene-11 «beta»-ol-3,17-dione 11beta-Hydroxyandrost-4-ene-3,17-dione
<b>Inchi:</b>	InChI=1S/C19H26O3/c1-18-8-7-12(20)9-11(18)3-4-13-14-5-6-16(22)19(14,2)10-15(21)17
<b>InchiKey:</b>	WSCUHXPGYUMQEX-UHFFFAOYSA-N
<b>Formula:</b>	C19H26O3
<b>SMILES:</b>	CC12CC(O)C3C(CCC4=CC(=O)CCC43C)C1CCC2=O
<b>Mol. weight [g/mol]:</b>	302.41
<b>CAS:</b>	382-44-5

## Physical Properties

Property code	Value	Unit	Source
gf	-96.47	kJ/mol	Joback Method
hf	-566.61	kJ/mol	Joback Method
hfus	20.49	kJ/mol	Joback Method
hvap	81.61	kJ/mol	Joback Method
log10ws	-3.94		Crippen Method
logp	3.058		Crippen Method
mcvol	239.840	ml/mol	McGowan Method
pc	2085.03	kPa	Joback Method
tb	905.53	K	Joback Method
tc	1151.14	K	Joback Method
tf	607.91	K	Joback Method
vc	0.900	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	880.55	J/molxK	905.53	Joback Method
cpg	905.20	J/molxK	946.47	Joback Method
cpg	930.17	J/molxK	987.40	Joback Method
cpg	955.78	J/molxK	1028.34	Joback Method

cpg	982.35	J/mol×K	1069.27	Joback Method
cpg	1010.19	J/mol×K	1110.21	Joback Method
cpg	1039.62	J/mol×K	1151.14	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C382445&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C382445&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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