

Barbital

Other names:

2,4,6(1H,3H,5H)-Pyrimidinetrione, 5,5-diethyl-
5,5-Diethyl-2,4,6(1H,3H,5H)-pyrimidinetrione
5,5-Diethylbarbituric acid
Barbitone
Barbituric acid, 5,5-diethyl-
Deba
Diemal
Diethylbarbitone
Diethylbarbituric acid
Diethylmalonylurea
Dormonal
Ethylbarbital
Hypnogene
Kyselina 5,5-diethylbarbiturova
Malonal
Metharbital M (nor)
NSC 31352
Sedeval
Uronal
Veroletten
Verolettin
Veronal
Vesperial

Inchi: InChI=1S/C8H12N2O3/c1-3-8(4-2)5(11)9-7(13)10-6(8)12/h3-4H2,1-2H3,(H2,9,10,11,12,13)1S

InchiKey: FTOAOBMCPZCFFF-UHFFFAOYSA-N

Formula: C8H12N2O3

SMILES: CCC1(CC)C(=O)NC(=O)NC1=O

Mol. weight [g/mol]: 184.19

CAS: 57-44-3

Physical Properties

Property code	Value	Unit	Source
chs	-4127.50	kJ/mol	NIST Webbook
chs	-4126.00	kJ/mol	NIST Webbook
gf	-156.91	kJ/mol	Joback Method
hf	-476.37	kJ/mol	Joback Method

h _{fus}	0.70	kJ/mol	Structural studies of cyclic ureas: 3. Enthalpy of formation of barbital
h _{sub}	117.30 ± 0.60	kJ/mol	NIST Webbook
h _{vap}	58.94	kJ/mol	Joback Method
log ₁₀ w _s	-1.41		Aqueous and cosolvent solubility data for drug-like organic compounds
log ₁₀ w _s	-2.40		Estimated Solubility Method
log ₁₀ w _s	-1.55		Aqueous Solubility Prediction Method
log _p	0.159		Crippen Method
m _{cvol}	137.390	ml/mol	McGowan Method
p _c	4000.70	kPa	Joback Method
r _{inpol}	1536.90		NIST Webbook
r _{inpol}	1469.00		NIST Webbook
r _{inpol}	1480.00		NIST Webbook
r _{inpol}	1480.00		NIST Webbook
r _{inpol}	1497.00		NIST Webbook
r _{inpol}	1457.00		NIST Webbook
r _{inpol}	1465.00		NIST Webbook
r _{inpol}	1462.00		NIST Webbook
r _{inpol}	1510.00		NIST Webbook
r _{inpol}	1465.00		NIST Webbook
r _{inpol}	1490.00		NIST Webbook
r _{inpol}	1467.00		NIST Webbook
r _{inpol}	1495.00		NIST Webbook
r _{inpol}	1473.00		NIST Webbook
r _{inpol}	1500.00		NIST Webbook
r _{inpol}	1495.00		NIST Webbook
r _{inpol}	1465.00		NIST Webbook
r _{inpol}	1497.00		NIST Webbook
r _{inpol}	1500.00		NIST Webbook
r _{inpol}	1480.00		NIST Webbook
r _{inpol}	1490.00		NIST Webbook
r _{inpol}	1490.00		NIST Webbook
r _{inpol}	1497.00		NIST Webbook
r _{inpol}	1482.00		NIST Webbook
r _{inpol}	1495.00		NIST Webbook
r _{inpol}	1500.00		NIST Webbook
r _{inpol}	1519.00		NIST Webbook
r _{inpol}	1496.00		NIST Webbook
r _{inpol}	1481.00		NIST Webbook
r _{inpol}	1500.00		NIST Webbook
r _{inpol}	1480.00		NIST Webbook

rmpol	1480.00		NIST Webbook
rmpol	1480.00		NIST Webbook
rmpol	1482.00		NIST Webbook
rmpol	1478.00		NIST Webbook
rmpol	1500.00		NIST Webbook
rmpol	1480.00		NIST Webbook
rmpol	1461.00		NIST Webbook
tb	702.79	K	Joback Method
tc	967.84	K	Joback Method
tf	463.00 ± 4.00	K	NIST Webbook
vc	0.509	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	387.82	J/mol×K	702.79	Joback Method
cpg	404.44	J/mol×K	746.96	Joback Method
cpg	420.27	J/mol×K	791.14	Joback Method
cpg	435.31	J/mol×K	835.31	Joback Method
cpg	449.56	J/mol×K	879.49	Joback Method
cpg	462.99	J/mol×K	923.66	Joback Method
cpg	475.62	J/mol×K	967.84	Joback Method
hfust	24.80	kJ/mol	462.00	NIST Webbook
hfust	24.98	kJ/mol	462.60	NIST Webbook
hsubt	113.90 ± 0.60	kJ/mol	366.00	NIST Webbook

Sources

- Estimated Solubility Method:** http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt
- Aqueous and cosolvent solubility data for drug-like organic compounds: McGowan Method:** <https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2751500/>
<http://link.springer.com/article/10.1007/BF02311772>
- NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C57443&Units=SI>
- Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>
- Structural studies of cyclic ureas: 3. Enthalpy of formation of barbital: Joback Method:** <https://www.doi.org/10.1016/j.jct.2009.06.018>
https://en.wikipedia.org/wiki/Joback_method
- Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

Legend

chs:	Standard solid enthalpy of combustion
cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hsub:	Enthalpy of sublimation at standard conditions
hsubt:	Enthalpy of sublimation at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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