

Ethylethynylcarbinol

Other names:	1-pentyn-3-ol Pent-1-yn-3-ol
Inchi:	InChI=1S/C5H8O/c1-3-5(6)4-2/h1,5-6H,4H2,2H3
InchiKey:	LBSKEFWQPNVWTP-UHFFFAOYSA-N
Formula:	C5H8O
SMILES:	C#CC(O)CC
Mol. weight [g/mol]:	84.12
CAS:	4187-86-4

Physical Properties

Property code	Value	Unit	Source
hfus	12.25	kJ/mol	Joback Method
ie	10.27	eV	Cheméo Relay (1.0)
logp	0.390		Crippen Method
mcvol	78.580	ml/mol	McGowan Method
tb	397.00	K	NIST Webbook

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	145.04	J/mol×K	395.66	Joback Method
cpg	151.97	J/mol×K	425.11	Joback Method
cpg	158.59	J/mol×K	454.56	Joback Method
cpg	164.90	J/mol×K	484.01	Joback Method
cpg	170.92	J/mol×K	513.46	Joback Method
cpg	176.67	J/mol×K	542.91	Joback Method
cpg	182.15	J/mol×K	572.36	Joback Method

Correlations

Information	Value
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Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.64802e+01
Coeff. B	-4.09810e+03
Coeff. C	-5.15160e+01
Temperature range (K), min.	304.15
Temperature range (K), max.	418.15

Sources

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

The Yaws Handbook of Vapor
Pressure:
NIST Webbook:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C4187864&Units=SI>

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
hfus:	Enthalpy of fusion at standard conditions
ie:	Ionization energy
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature

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