

# Alpha-phenylethyl n-butyl ether

<b>Other names:</b>	(1-Butoxyethyl)benzene
<b>Inchi:</b>	InChI=1S/C12H18O/c1-3-4-10-13-11(2)12-8-6-5-7-9-12/h5-9,11H,3-4,10H2,1-2H3
<b>InchiKey:</b>	AQGPVCBVZAHVBN-UHFFFAOYSA-N
<b>Formula:</b>	C12H18O
<b>SMILES:</b>	CCCCOC(C)c1ccccc1
<b>Mol. weight [g/mol]:</b>	178.27
<b>CAS:</b>	4157-77-1

## Physical Properties

Property code	Value	Unit	Source
gf	55.13	kJ/mol	Joback Method
hf	-191.98	kJ/mol	Joback Method
hfus	18.54	kJ/mol	Joback Method
hvap	59.80 ± 0.30	kJ/mol	NIST Webbook
log10ws	-3.49		Crippen Method
logp	3.564		Crippen Method
mcvol	162.050	ml/mol	McGowan Method
pc	2384.19	kPa	Joback Method
tb	522.62	K	Joback Method
tc	724.59	K	Joback Method
tf	258.65	K	Joback Method
vc	0.612	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	372.16	J/mol×K	522.62	Joback Method
cpg	389.01	J/mol×K	556.28	Joback Method
cpg	404.99	J/mol×K	589.94	Joback Method
cpg	420.13	J/mol×K	623.60	Joback Method
cpg	434.45	J/mol×K	657.27	Joback Method
cpg	447.97	J/mol×K	690.93	Joback Method
cpg	460.72	J/mol×K	724.59	Joback Method
dvisc	0.0039484	Paxs	258.65	Joback Method

dvisc	0.0015764	Paxs	302.64	Joback Method
dvisc	0.0007945	Paxs	346.64	Joback Method
dvisc	0.0004673	Paxs	390.63	Joback Method
dvisc	0.0003060	Paxs	434.63	Joback Method
dvisc	0.0002166	Paxs	478.62	Joback Method
dvisc	0.0001625	Paxs	522.62	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4157771&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4157771&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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