

# Dihydrojasmone

<b>Other names:</b>	2-Cyclopenten-1-one, 3-methyl-2-pentyl-Jasmone, dihydro- 3-Methyl-2-(n-pentanyl)-2-cyclopenten-1-one 2-Pentyl-3-methyl-2-cyclopenten-1-one 2-n-Pentyl-3-methyl-2-cyclopenten-1-one 3-methyl-2-pentylcyclopent-2-enone
<b>Inchi:</b>	InChI=1S/C11H18O/c1-3-4-5-6-10-9(2)7-8-11(10)12/h3-8H2,1-2H3
<b>InchiKey:</b>	YCIXWYOBMVNGTB-UHFFFAOYSA-N
<b>Formula:</b>	C11H18O
<b>SMILES:</b>	<chem>CCCCCC1=C(C)CCC1=O</chem>
<b>Mol. weight [g/mol]:</b>	166.26
<b>CAS:</b>	1128-08-1

## Physical Properties

Property code	Value	Unit	Source
gf	-25.89	kJ/mol	Joback Method
hf	-292.41	kJ/mol	Joback Method
hfus	17.06	kJ/mol	Joback Method
hvap	46.51	kJ/mol	Joback Method
log10ws	-3.46		Crippen Method
logp	3.246		Crippen Method
mcvol	152.260	ml/mol	McGowan Method
pc	2450.74	kPa	Joback Method
rinpol	1374.00		NIST Webbook
rinpol	1400.00		NIST Webbook
rinpol	1362.40		NIST Webbook
ripol	1842.00		NIST Webbook
ripol	1892.00		NIST Webbook
tb	547.97	K	Joback Method
tc	755.00	K	Joback Method
tf	322.89	K	Joback Method
vc	0.587	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	367.24	J/mol×K	547.97	Joback Method
cpg	383.69	J/mol×K	582.48	Joback Method
cpg	399.40	J/mol×K	616.98	Joback Method
cpg	414.36	J/mol×K	651.49	Joback Method
cpg	428.57	J/mol×K	685.99	Joback Method
cpg	442.06	J/mol×K	720.50	Joback Method
cpg	454.82	J/mol×K	755.00	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	416.70	K	2.90	NIST Webbook
tbrp	389.20	K	1.60	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1128081&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1128081&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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