

# Diacetamide

<b>Other names:</b>	CH <sub>3</sub> C(O)NHC(O)CH <sub>3</sub> Bisacetylamine Acetamide, N-acetyl- Dicetylamine N-Acetylacetamide Diacetylamine
<b>Inchi:</b>	InChI=1S/C4H7NO2/c1-3(6)5-4(2)7/h1-2H3,(H,5,6,7)
<b>InchiKey:</b>	ZSBDPRIWBYHIAF-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>4</sub> H <sub>7</sub> NO <sub>2</sub>
<b>SMILES:</b>	CC(=O)NC(C)=O
<b>Mol. weight [g/mol]:</b>	101.10
<b>CAS:</b>	625-77-4

## Physical Properties

Property code	Value	Unit	Source
chs	-2087.52	kJ/mol	NIST Webbook
gf	-185.65	kJ/mol	Joback Method
hf	-297.58	kJ/mol	Joback Method
hfs	-500.80	kJ/mol	NIST Webbook
hfus	14.41	kJ/mol	Joback Method
hsub	73.20 ± 0.80	kJ/mol	NIST Webbook
hsub	73.20 ± 0.80	kJ/mol	NIST Webbook
hvap	44.43	kJ/mol	Joback Method
log10ws	-0.24		Crippen Method
logp	-0.331		Crippen Method
mcvol	80.340	ml/mol	McGowan Method
pc	4672.09	kPa	Joback Method
tb	496.70	K	NIST Webbook
tc	645.66	K	Joback Method
tf	287.36	K	Joback Method
vc	0.306	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	155.41	J/mol×K	448.83	Joback Method
cpg	163.03	J/mol×K	481.64	Joback Method
cpg	170.29	J/mol×K	514.44	Joback Method
cpg	177.20	J/mol×K	547.25	Joback Method
cpg	183.77	J/mol×K	580.05	Joback Method
cpg	190.02	J/mol×K	612.86	Joback Method
cpg	195.93	J/mol×K	645.66	Joback Method
hvapt	59.70	kJ/mol	432.00	NIST Webbook
hvapt	64.60	kJ/mol	419.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C625774&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C625774&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/40-656-9/Diacetamide.pdf>

Generated by Cheméo on 2024-08-08 07:40:47.094475508 +0000 UTC m=+1792716.341580855.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.