

Dichloroacetic acid, 2-butyl ester

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| Other names: | Acetic acid, dichloro, 1-methylpropyl ester |
| Inchi: | InChI=1S/C6H10Cl2O2/c1-3-4(2)10-6(9)5(7)8/h4-5H,3H2,1-2H3 |
| InchiKey: | IQYVZHJFEAIITM-UHFFFAOYSA-N |
| Formula: | C6H10Cl2O2 |
| SMILES: | CCC(C)OC(=O)C(Cl)Cl |
| Mol. weight [g/mol]: | 185.05 |
| CAS: | 62223-12-5 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | -263.02 | kJ/mol | Joback Method |
| hf | -454.01 | kJ/mol | Joback Method |
| hfus | 15.43 | kJ/mol | Joback Method |
| hvap | 46.10 | kJ/mol | Joback Method |
| log10ws | -2.22 | | Crippen Method |
| logp | 2.132 | | Crippen Method |
| mcvol | 127.320 | ml/mol | McGowan Method |
| pc | 3065.95 | kPa | Joback Method |
| rinpol | 1015.00 | | NIST Webbook |
| ripol | 1427.00 | | NIST Webbook |
| tb | 486.95 | K | Joback Method |
| tc | 685.84 | K | Joback Method |
| tf | 259.38 | K | Joback Method |
| vc | 0.481 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 249.45 | J/molxK | 486.95 | Joback Method |
| cpg | 294.25 | J/molxK | 652.69 | Joback Method |
| cpg | 286.16 | J/molxK | 619.54 | Joback Method |
| cpg | 277.63 | J/molxK | 586.39 | Joback Method |
| cpg | 268.68 | J/molxK | 553.25 | Joback Method |
| cpg | 259.28 | J/molxK | 520.10 | Joback Method |

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|-------|-----------|---------|--------|---------------|
| cpg | 301.92 | J/molxK | 685.84 | Joback Method |
| dvisc | 0.0002759 | Paxs | 486.95 | Joback Method |
| dvisc | 0.0003681 | Paxs | 449.02 | Joback Method |
| dvisc | 0.0005180 | Paxs | 411.09 | Joback Method |
| dvisc | 0.0007812 | Paxs | 373.16 | Joback Method |
| dvisc | 0.0012930 | Paxs | 335.24 | Joback Method |
| dvisc | 0.0024337 | Paxs | 297.31 | Joback Method |
| dvisc | 0.0055114 | Paxs | 259.38 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C62223125&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| ripol: | Non-polar retention indices |
| ripol: | Polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

Latest version available from:

<https://www.chemeo.com/cid/40-680-2/Dichloroacetic-acid-2-butyl-ester.pdf>

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