

Benzene, beta-deuteroethynyl-

Inchi:	InChI=1S/C8H6/c1-2-8-6-4-3-5-7-8/h1,3-7H/i1D
InchiKey:	UEXCJVNBTNXOEH-MICDWDOJSA-N
Formula:	C8H5D
SMILES:	C#Cc1ccccc1
Mol. weight [g/mol]:	103.14

Physical Properties

Property code	Value	Unit	Source
gf	351.96	kJ/mol	Joback Method
hf	319.98	kJ/mol	Joback Method
hfus	13.49	kJ/mol	Joback Method
hvap	35.54	kJ/mol	Joback Method
log10ws	-2.17		Crippen Method
logp	1.668		Crippen Method
mcvol	91.220	ml/mol	McGowan Method
pc	4403.25	kPa	Joback Method
tb	399.24	K	Joback Method
tc	625.47	K	Joback Method
tf	253.31	K	Joback Method
vc	0.338	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	149.68	J/mol×K	399.24	Joback Method
cpg	160.67	J/mol×K	436.95	Joback Method
cpg	170.89	J/mol×K	474.65	Joback Method
cpg	180.38	J/mol×K	512.36	Joback Method
cpg	189.18	J/mol×K	550.06	Joback Method
cpg	197.33	J/mol×K	587.77	Joback Method
cpg	204.88	J/mol×K	625.47	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=B6010137&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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