

# 1,2,4-Metheno, 1H-indene

<b>Inchi:</b>	InChI=1S/C10H6/c1-2-5-7-4-8-9(5)10(8)6(7)3-1/h1-4,9-10H
<b>InchiKey:</b>	ZKOZJNMUXZLEPP-UHFFFAOYSA-N
<b>Formula:</b>	C10H6
<b>SMILES:</b>	C1=C2C3c4cccc(c41)C23
<b>Mol. weight [g/mol]:</b>	126.15

## Physical Properties

Property code	Value	Unit	Source
gf	397.26	kJ/mol	Joback Method
hf	279.71	kJ/mol	Joback Method
hfus	19.59	kJ/mol	Joback Method
hvap	41.60	kJ/mol	Joback Method
log10ws	-2.67		Crippen Method
logp	2.278		Crippen Method
mcvol	91.120	ml/mol	McGowan Method
pc	4119.70	kPa	Joback Method
rinpol	1365.00		NIST Webbook
rinpol	1365.00		NIST Webbook
tb	472.52	K	Joback Method
tc	693.55	K	Joback Method
tf	342.86	K	Joback Method
vc	0.386	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	200.99	J/mol×K	472.52	Joback Method
cpg	211.85	J/mol×K	509.36	Joback Method
cpg	221.46	J/mol×K	546.20	Joback Method
cpg	229.97	J/mol×K	583.03	Joback Method
cpg	237.54	J/mol×K	619.87	Joback Method
cpg	244.35	J/mol×K	656.71	Joback Method
cpg	250.54	J/mol×K	693.55	Joback Method
dvisc	0.0010039	Paxs	342.86	Joback Method

dvisc	0.0012637	Paxs	364.47	Joback Method
dvisc	0.0015503	Paxs	386.08	Joback Method
dvisc	0.0018611	Paxs	407.69	Joback Method
dvisc	0.0021934	Paxs	429.30	Joback Method
dvisc	0.0025447	Paxs	450.91	Joback Method
dvisc	0.0029124	Paxs	472.52	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R413402&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R413402&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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