

# Propyl sulfide

<b>Other names:</b>	(n-C3H7)2S 1,1'-Thiobispropane 4-THIAHEPTANE DIPROPYL THIOETHER Di-n-propyl thioether Dipropyl sulfide Dipropyl sulphide NSC 78429 Propane, 1,1'-thiobis- Propyl monosulfide Sulfide, n-propyl- di-n-Propyl sulfide di-n-propylsulfide n-Propyl sulfide
<b>Inchi:</b>	InChI=1S/C6H14S/c1-3-5-7-6-4-2/h3-6H2,1-2H3
<b>InchiKey:</b>	ZERULLAPCVRMCO-UHFFFAOYSA-N
<b>Formula:</b>	C6H14S
<b>SMILES:</b>	CCCSCCC
<b>Mol. weight [g/mol]:</b>	118.24
<b>CAS:</b>	111-47-7

## Physical Properties

Property code	Value	Unit	Source
affp	864.70	kJ/mol	NIST Webbook
basg	834.90	kJ/mol	NIST Webbook
chl	-4794.28 ± 0.75	kJ/mol	NIST Webbook
gf	32.76	kJ/mol	Joback Method
hf	-125.20	kJ/mol	NIST Webbook
hfl	-169.80 ± 0.88	kJ/mol	NIST Webbook
hfus	15.43	kJ/mol	Joback Method
hvap	39.50	kJ/mol	NIST Webbook
hvap	44.20	kJ/mol	NIST Webbook
hvap	44.50	kJ/mol	NIST Webbook
hvap	44.23	kJ/mol	NIST Webbook
hvap	44.70 ± 0.80	kJ/mol	NIST Webbook
hvap	44.60	kJ/mol	NIST Webbook
ie	8.34	eV	NIST Webbook

ie	8.30 ± 0.02	eV	NIST Webbook
ie	8.34	eV	NIST Webbook
ie	8.45 ± 0.05	eV	NIST Webbook
ie	10.90	eV	NIST Webbook
log10ws	-2.58		Aqueous Solubility Prediction Method
log10ws	-2.58		Estimated Solubility Method
logp	2.540		Crippen Method
mcvol	111.750	ml/mol	McGowan Method
pc	3166.83	kPa	Joback Method
rinpol	878.00		NIST Webbook
rinpol	872.00		NIST Webbook
rinpol	894.00		NIST Webbook
rinpol	883.00		NIST Webbook
rinpol	878.00		NIST Webbook
rinpol	887.00		NIST Webbook
rinpol	885.00		NIST Webbook
rinpol	878.00		NIST Webbook
rinpol	877.00		NIST Webbook
rinpol	890.00		NIST Webbook
rinpol	897.00		NIST Webbook
rinpol	891.00		NIST Webbook
rinpol	894.00		NIST Webbook
rinpol	878.00		NIST Webbook
rinpol	888.00		NIST Webbook
rinpol	879.00		NIST Webbook
rinpol	888.00		NIST Webbook
rinpol	883.00		NIST Webbook
rinpol	877.00		NIST Webbook
rinpol	890.00		NIST Webbook
rinpol	878.00		NIST Webbook
rinpol	891.20		NIST Webbook
rinpol	887.00		NIST Webbook
rinpol	891.20		NIST Webbook
rinpol	896.70		NIST Webbook
rinpol	884.80		NIST Webbook
rinpol	887.40		NIST Webbook
rinpol	882.00		NIST Webbook
rinpol	887.00		NIST Webbook
rinpol	888.00		NIST Webbook
rinpol	888.00		NIST Webbook
rinpol	890.00		NIST Webbook
rinpol	882.00		NIST Webbook
rinpol	888.00		NIST Webbook

ripol	888.00		NIST Webbook
ripol	1069.00		NIST Webbook
ripol	1101.00		NIST Webbook
ripol	1093.00		NIST Webbook
ripol	1091.00		NIST Webbook
ripol	1080.50		NIST Webbook
ripol	1069.00		NIST Webbook
sl	338.28	J/molxK	NIST Webbook
tb	416.00 ± 0.40	K	NIST Webbook
tb	414.70 ± 0.50	K	NIST Webbook
tb	415.60	K	NIST Webbook
tb	414.65 ± 0.50	K	NIST Webbook
tb	415.60 ± 0.50	K	NIST Webbook
tb	412.00 ± 2.00	K	NIST Webbook
tb	415.85 ± 0.20	K	NIST Webbook
tb	414.00 ± 0.50	K	NIST Webbook
tb	414.00 ± 0.50	K	NIST Webbook
tb	416.00	K	NIST Webbook
tb	415.00 ± 2.00	K	NIST Webbook
tc	594.29	K	Joback Method
tf	170.65	K	Aqueous Solubility Prediction Method
tt	170.44 ± 0.03	K	NIST Webbook
tt	170.43 ± 0.05	K	NIST Webbook
vc	0.425	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	268.89	J/molxK	594.29	Joback Method
cpg	206.46	J/molxK	405.46	Joback Method
cpg	259.54	J/molxK	562.82	Joback Method
cpg	249.77	J/molxK	531.34	Joback Method
cpg	239.58	J/molxK	499.87	Joback Method
cpg	228.97	J/molxK	468.40	Joback Method
cpg	217.94	J/molxK	436.93	Joback Method
cpl	225.48	J/molxK	298.15	NIST Webbook
cpl	237.90	J/molxK	300.00	NIST Webbook
hfust	12.14	kJ/mol	170.40	NIST Webbook
hfust	12.13	kJ/mol	170.40	NIST Webbook
hfust	12.14	kJ/mol	170.44	NIST Webbook

hvapt	36.60	kJ/mol	416.00	NIST Webbook
hvapt	40.60	kJ/mol	390.00	NIST Webbook
hvapt	42.90	kJ/mol	362.00	NIST Webbook
sfust	71.24	J/molxK	170.44	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.48928e+01
Coeff. B	-3.68038e+03
Coeff. C	-5.67920e+01
Temperature range (K), min.	308.78
Temperature range (K), max.	440.91

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	1.02676e+02
Coeff. B	-8.82408e+03
Coeff. C	-1.29840e+01
Coeff. D	8.43544e-06
Temperature range (K), min.	270.15
Temperature range (K), max.	609.73

## Sources

<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C111477&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C111477&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1836">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1836</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol1836.mol">https://www.thermo.com/files/research/kdb/mol/mol1836.mol</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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