

# Pentanedioic acid, 3-oxo-, dimethyl ester

<b>Other names:</b>	Acetone dicarboxylic acid, dimethyl ester Dimethyl acetonedicarboxylate Dimethyl 1,3-acetonedicarboxylate Dimethyl «beta»-ketoglutarate Dimethyl «beta»-oxoglutarate Dimethyl 3-oxopentanedioate Glutaric acid, 3-oxo-, dimethyl ester Dimethyl 3-oxoglutarate NSC 4677 Pentanedioic acid, 3-oxo-, 1,5-dimethyl ester
<b>Inchi:</b>	InChI=1S/C7H10O5/c1-11-6(9)3-5(8)4-7(10)12-2/h3-4H2,1-2H3
<b>InchiKey:</b>	RNJOKCPFLQMDEC-UHFFFAOYSA-N
<b>Formula:</b>	C7H10O5
<b>SMILES:</b>	<chem>COC(=O)CC(=O)CC(=O)OC</chem>
<b>Mol. weight [g/mol]:</b>	174.15
<b>CAS:</b>	1830-54-2

## Physical Properties

Property code	Value	Unit	Source
gf	-588.70	kJ/mol	Joback Method
hf	-789.99	kJ/mol	Joback Method
hfus	21.06	kJ/mol	Joback Method
hvap	56.23	kJ/mol	Joback Method
log10ws	0.24		Crippen Method
logp	-0.318		Crippen Method
mcvol	125.940	ml/mol	McGowan Method
pc	3322.01	kPa	Joback Method
tb	566.01	K	Joback Method
tc	760.26	K	Joback Method
tf	362.90	K	Joback Method
vc	0.481	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	292.09	J/mol×K	566.01	Joback Method
cpg	301.83	J/mol×K	598.38	Joback Method
cpg	311.16	J/mol×K	630.76	Joback Method
cpg	320.07	J/mol×K	663.13	Joback Method
cpg	328.54	J/mol×K	695.51	Joback Method
cpg	336.57	J/mol×K	727.88	Joback Method
cpg	344.14	J/mol×K	760.26	Joback Method
dvisc	0.0017994	Paxs	362.90	Joback Method
dvisc	0.0011370	Paxs	396.75	Joback Method
dvisc	0.0007723	Paxs	430.60	Joback Method
dvisc	0.0005550	Paxs	464.46	Joback Method
dvisc	0.0004171	Paxs	498.31	Joback Method
dvisc	0.0003251	Paxs	532.16	Joback Method
dvisc	0.0002610	Paxs	566.01	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	423.20	K	3.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1830542&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1830542&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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