

# Sebacic acid, di(3-ethylphenyl) ester

<b>Inchi:</b>	InChI=1S/C26H34O4/c1-3-21-13-11-15-23(19-21)29-25(27)17-9-7-5-6-8-10-18-26(28)30
<b>InchiKey:</b>	IPGFBWALSOCWRT-UHFFFAOYSA-N
<b>Formula:</b>	C26H34O4
<b>SMILES:</b>	CCc1cccc(OC(=O)CCCCCCCC(=O)Oc2cccc(CC)c2)c1
<b>Mol. weight [g/mol]:</b>	410.55

## Physical Properties

Property code	Value	Unit	Source
gf	-94.24	kJ/mol	Joback Method
hf	-619.45	kJ/mol	Joback Method
hfus	55.97	kJ/mol	Joback Method
hvap	97.66	kJ/mol	Joback Method
log10ws	-7.87		Crippen Method
logp	6.443		Crippen Method
mcvol	344.560	ml/mol	McGowan Method
pc	1118.56	kPa	Joback Method
rinqol	3252.00		NIST Webbook
tb	1010.18	K	Joback Method
tc	1238.38	K	Joback Method
tf	604.98	K	Joback Method
vc	1.323	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1136.73	J/molxK	1010.18	Joback Method
cpg	1194.43	J/molxK	1200.35	Joback Method
cpg	1185.67	J/molxK	1162.32	Joback Method
cpg	1175.57	J/molxK	1124.28	Joback Method
cpg	1164.09	J/molxK	1086.25	Joback Method
cpg	1151.16	J/molxK	1048.21	Joback Method
cpg	1201.91	J/molxK	1238.38	Joback Method
dvisc	0.0000242	Paxs	1010.18	Joback Method
dvisc	0.0000310	Paxs	942.65	Joback Method

dvisc	0.0000411	Paxs	875.11	Joback Method
dvisc	0.0000572	Paxs	807.58	Joback Method
dvisc	0.0000846	Paxs	740.05	Joback Method
dvisc	0.0001352	Paxs	672.51	Joback Method
dvisc	0.0002402	Paxs	604.98	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U355248&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U355248&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/40-934-0/Sebacic-acid-di-3-ethylphenyl-ester.pdf>

Generated by Cheméo on 2024-04-26 14:47:53.364667007 +0000 UTC m=+16432122.285244322.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.