

# Tetrahydrofuran, 3-chloro-2-(4-chlorobutyloxy)

<b>Other names:</b>	2-(4-Chloro-butoxy)-3-chloro-tetrahydro-furan
<b>Inchi:</b>	InChI=1S/C8H14Cl2O2/c9-4-1-2-5-11-8-7(10)3-6-12-8/h7-8H,1-6H2
<b>InchiKey:</b>	LWNIBOOTGWNIBR-UHFFFAOYSA-N
<b>Formula:</b>	C8H14Cl2O2
<b>SMILES:</b>	C1CCCCOC1OCCC1Cl
<b>Mol. weight [g/mol]:</b>	213.10

## Physical Properties

Property code	Value	Unit	Source
gf	-169.66	kJ/mol	Joback Method
hf	-464.01	kJ/mol	Joback Method
hfus	29.04	kJ/mol	Joback Method
hvap	49.04	kJ/mol	Joback Method
log10ws	-2.27		Crippen Method
logp	2.376		Crippen Method
mcvol	148.940	ml/mol	McGowan Method
pc	2659.77	kPa	Joback Method
rinpol	1415.00		NIST Webbook
rinpol	1415.00		NIST Webbook
tb	517.28	K	Joback Method
tc	720.80	K	Joback Method
tf	295.22	K	Joback Method
vc	0.560	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	332.03	J/mol×K	517.28	Joback Method
cpg	398.19	J/mol×K	686.88	Joback Method
cpg	386.37	J/mol×K	652.96	Joback Method
cpg	373.86	J/mol×K	619.04	Joback Method
cpg	360.64	J/mol×K	585.12	Joback Method
cpg	346.70	J/mol×K	551.20	Joback Method
cpg	409.33	J/mol×K	720.80	Joback Method

dvisc	0.0003509	Paxs	517.28	Joback Method
dvisc	0.0004335	Paxs	480.27	Joback Method
dvisc	0.0005547	Paxs	443.26	Joback Method
dvisc	0.0007425	Paxs	406.25	Joback Method
dvisc	0.0010537	Paxs	369.24	Joback Method
dvisc	0.0016165	Paxs	332.23	Joback Method
dvisc	0.0027608	Paxs	295.22	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R91280&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R91280&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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