

# Pinonic acid

**Other names:**

(-)-cis-pinonic acid  
2-(3-Acetyl-2,2-dimethylcyclobutyl)acetic acid  
Cyclobutaneacetic acid, 3-acetyl-2,2-dimethyl-  
NSC 29469  
cis-3-acetyl-2,2-dimethylcyclobutaneacetic acid  
cis-3-acetyl-2,2-dimethylcyclobutylacetic acid  
cis-pinonic acid  
dl-cis-pinonic acid

**Inchi:**

InChI=1S/C10H16O3/c1-6(11)8-4-7(5-9(12)13)10(8,2)3/h7-8H,4-5H2,1-3H3,(H,12,13)/t7

**InchiKey:**

SIZDUQQDBXJXLQ-JGVFFNPUSA-N

**Formula:**

C10H16O3

**SMILES:**

CC(=O)C1CC(CC(=O)O)C1(C)C

**Mol. weight [g/mol]:**

184.23

**CAS:**

473-72-3

## Physical Properties

Property code	Value	Unit	Source
gf	-333.60	kJ/mol	Joback Method
hf	-585.92	kJ/mol	Joback Method
hfus	20.82	kJ/mol	Joback Method
hvap	66.34	kJ/mol	Joback Method
log10ws	-1.56		Crippen Method
logp	1.712		Crippen Method
mcvol	149.910	ml/mol	McGowan Method
pc	2999.15	kPa	Joback Method
rinpol	1427.00		NIST Webbook
tb	630.03	K	Joback Method
tc	826.05	K	Joback Method
tf	392.98	K	Joback Method
vc	0.572	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	409.51	J/mol×K	630.03	Joback Method
cpg	422.36	J/mol×K	662.70	Joback Method
cpg	434.58	J/mol×K	695.37	Joback Method
cpg	446.27	J/mol×K	728.04	Joback Method
cpg	457.50	J/mol×K	760.71	Joback Method
cpg	468.37	J/mol×K	793.38	Joback Method
cpg	478.95	J/mol×K	826.05	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Surface Tensions and Densities of Oxalic, Malonic, Succinic, Maleic, Malic, and Glutaric Acids:</b>	<a href="https://www.doi.org/10.1021/je050366x">https://www.doi.org/10.1021/je050366x</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C473723&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C473723&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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