

Ethanone, 1-(3-fluorophenyl)-

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|-----------------------------|---|
| Other names: | Acetophenone, 3'-fluoro- m-Fluoroacetophenone 3-Fluoroacetophenone 3'-Fluoroacetophenone |
| Inchi: | InChI=1S/C8H7FO/c1-6(10)7-3-2-4-8(9)5-7/h2-5H,1H3 |
| InchiKey: | HCEKGPAAHZCYRBZ-UHFFFAOYSA-N |
| Formula: | C8H7FO |
| SMILES: | CC(=O)c1cccc(F)c1 |
| Mol. weight [g/mol]: | 138.14 |
| CAS: | 455-36-7 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|-------------|----------------------|----------------|
| affp | 845.70 | kJ/mol | NIST Webbook |
| basg | 813.80 | kJ/mol | NIST Webbook |
| ea | 0.58 ± 0.03 | eV | NIST Webbook |
| gf | -204.47 | kJ/mol | Joback Method |
| hf | -292.08 | kJ/mol | Joback Method |
| hfus | 14.81 | kJ/mol | Joback Method |
| hvap | 42.27 | kJ/mol | Joback Method |
| ie | 9.80 ± 0.10 | eV | NIST Webbook |
| log10ws | -2.46 | | Crippen Method |
| logp | 2.028 | | Crippen Method |
| mcvol | 103.160 | ml/mol | McGowan Method |
| pc | 3664.21 | kPa | Joback Method |
| tb | 467.24 | K | Joback Method |
| tc | 678.42 | K | Joback Method |
| tf | 269.38 | K | Joback Method |
| vc | 0.400 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 200.90 | J/molxK | 467.24 | Joback Method |

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|-----|--------|---------|--------|---------------|
| cpg | 211.62 | J/mol×K | 502.44 | Joback Method |
| cpg | 221.72 | J/mol×K | 537.63 | Joback Method |
| cpg | 231.22 | J/mol×K | 572.83 | Joback Method |
| cpg | 240.13 | J/mol×K | 608.03 | Joback Method |
| cpg | 248.49 | J/mol×K | 643.23 | Joback Method |
| cpg | 256.32 | J/mol×K | 678.42 | Joback Method |

Pressure Dependent Properties

| Property code | Value | Unit | Pressure [kPa] | Source |
|---------------|--------|------|----------------|--------------|
| tbrp | 354.20 | K | 1.00 | NIST Webbook |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C455367&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| affp: | Proton affinity |
| basg: | Gas basicity |
| cpg: | Ideal gas heat capacity |
| ea: | Electron affinity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| ie: | Ionization energy |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |

| | |
|--------------|-----------------------------------|
| tb: | Normal Boiling Point Temperature |
| tbrp: | Boiling point at reduced pressure |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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<https://www.chemeo.com/cid/41-089-8/Ethanone-1-3-fluorophenyl.pdf>

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