

# Fumaric acid, 2,2-dichloroethyl pentadecyl ester

Inchi:	InChI=1S/C21H36Cl2O4/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-17-26-20(24)15-16-21(25)27
InchiKey:	MWXYEQDJMAXBGQ-FOCLMDBBSA-N
Formula:	C21H36Cl2O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C=CC(=O)OCC(Cl)Cl
Mol. weight [g/mol]:	423.41

## Physical Properties

Property code	Value	Unit	Source
gf	-287.98	kJ/mol	Joback Method
hf	-885.91	kJ/mol	Joback Method
hfus	60.79	kJ/mol	Joback Method
hvap	88.99	kJ/mol	Joback Method
log10ws	-7.10		Crippen Method
logp	6.524		Crippen Method
mvol	341.810	ml/mol	McGowan Method
pc	1003.98	kPa	Joback Method
rinpol	2827.00		NIST Webbook
tb	911.04	K	Joback Method
tc	1115.57	K	Joback Method
tf	510.51	K	Joback Method
vc	1.331	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1064.07	J/molxK	911.04	Joback Method
cpg	1134.70	J/molxK	1081.48	Joback Method
cpg	1122.68	J/molxK	1047.39	Joback Method
cpg	1109.64	J/molxK	1013.30	Joback Method
cpg	1095.56	J/molxK	979.22	Joback Method
cpg	1080.38	J/molxK	945.13	Joback Method
cpg	1145.76	J/molxK	1115.57	Joback Method
dvisc	0.0000264	Paxs	911.04	Joback Method
dvisc	0.0000354	Paxs	844.28	Joback Method

dvisc	0.0000499	Paxs	777.53	Joback Method
dvisc	0.0000750	Paxs	710.77	Joback Method
dvisc	0.0001228	Paxs	644.02	Joback Method
dvisc	0.0002254	Paxs	577.26	Joback Method
dvisc	0.0004848	Paxs	510.51	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348582&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348582&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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