

# N'-(3-methyl-phenyl)-N,N-dimethyl-acetamidine

Inchi:	lnChI=1S/C11H16N2/c1-9-6-5-7-11(8-9)12-10(2)13(3)4/h5-8H,1-4H3/b12-10+
InchiKey:	UQSHARSOYYHIAZ-ZRDIBKRKSA-N
Formula:	C11H16N2
SMILES:	CC(=Nc1cccc(C)c1)N(C)C
Mol. weight [g/mol]:	176.26

## Physical Properties

Property code	Value	Unit	Source
hf	94.65	kJ/mol	Joback Method
hvap	48.45	kJ/mol	Joback Method
log10ws	-2.42		Crippen Method
logp	2.607		Crippen Method
mcvol	157.750	ml/mol	McGowan Method
pc	2356.49	kPa	Joback Method
rinpol	1547.00		NIST Webbook
tb	571.74	K	Joback Method
tc	793.72	K	Joback Method

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R153318&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R153318&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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