

# Androstane-3,17-dione, (5 $\alpha$ ), 2,2,4,4,16,16-hexadeuterio-

<b>Inchi:</b>	InChI=1S/C19H28O2/c1-18-9-7-13(20)11-12(18)3-4-14-15-5-6-17(21)19(15,2)10-8-16(14)
<b>InchiKey:</b>	RAJWOBJTTGJROA-FCLGVVOLSA-N
<b>Formula:</b>	C19H22D6O2
<b>SMILES:</b>	CC12CCC3C(CCC4CC(=O)CCC43C)C1CCC2=O
<b>Mol. weight [g/mol]:</b>	294.46

## Physical Properties

Property code	Value	Unit	Source
gf	20.02	kJ/mol	Joback Method
hf	-460.69	kJ/mol	Joback Method
hfus	15.57	kJ/mol	Joback Method
hvap	63.97	kJ/mol	Joback Method
log10ws	-4.46		Crippen Method
logp	4.167		Crippen Method
mcvol	238.270	ml/mol	McGowan Method
pc	1885.44	kPa	Joback Method
tb	809.21	K	Joback Method
tc	1073.00	K	Joback Method
tf	533.81	K	Joback Method
vc	0.895	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	834.08	J/mol×K	809.21	Joback Method
cpg	862.01	J/mol×K	853.18	Joback Method
cpg	889.40	J/mol×K	897.14	Joback Method
cpg	916.66	J/mol×K	941.11	Joback Method
cpg	944.17	J/mol×K	985.07	Joback Method
cpg	972.33	J/mol×K	1029.04	Joback Method
cpg	1001.55	J/mol×K	1073.00	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=B6009861&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=B6009861&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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