

# Benzeneethanamine, N,N,«alpha»-trimethyl-

<b>Other names:</b>	Phenethylamine, N,N,«alpha»-trimethyl-Dimethylamphetamine N,N,«alpha»-Trimethylphenethylamine 2-Dimethylamino-1-phenylpropane 1-Phenyl-2-dimethylamino-propan «alpha»-Phenyl-«beta»-dimethylaminopropane 1-Phenyl-2-dimethylaminopropane N,N-Dimethylamphetamine
<b>Inchi:</b>	InChI=1S/C11H17N/c1-10(12(2)3)9-11-7-5-4-6-8-11/h4-8,10H,9H2,1-3H3
<b>InchiKey:</b>	OBDSVYOSYSKVMX-UHFFFAOYSA-N
<b>Formula:</b>	C11H17N
<b>SMILES:</b>	CC(Cc1ccccc1)N(C)C
<b>Mol. weight [g/mol]:</b>	163.26
<b>CAS:</b>	4075-96-1

## Physical Properties

Property code	Value	Unit	Source
gf	262.49	kJ/mol	Joback Method
hf	28.41	kJ/mol	Joback Method
hfus	17.79	kJ/mol	Joback Method
hvap	44.01	kJ/mol	Joback Method
log10ws	-2.21		Crippen Method
logp	2.179		Crippen Method
mcvol	152.070	ml/mol	McGowan Method
pc	2665.27	kPa	Joback Method
rinpol	1236.00		NIST Webbook
rinpol	1233.00		NIST Webbook
rinpol	1243.00		NIST Webbook
rinpol	1236.00		NIST Webbook
rinpol	1243.00		NIST Webbook
rinpol	1251.00		NIST Webbook
rinpol	1251.00		NIST Webbook
ripol	1568.00		NIST Webbook
ripol	1568.00		NIST Webbook
tb	489.76	K	Joback Method
tc	693.60	K	Joback Method
tf	257.62	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	333.54	J/mol×K	489.76	Joback Method
cpg	350.82	J/mol×K	523.73	Joback Method
cpg	367.10	J/mol×K	557.71	Joback Method
cpg	382.41	J/mol×K	591.68	Joback Method
cpg	396.81	J/mol×K	625.65	Joback Method
cpg	410.33	J/mol×K	659.62	Joback Method
cpg	423.02	J/mol×K	693.60	Joback Method

## Sources

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C4075961&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/41-292-2/Benzeneethanamine-N-N-alpha-trimethyl.pdf>

Generated by Cheméo on 2022-12-05 11:42:23.660356884 +0000 UTC m=+236906.397222578.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.