

# Heptane, 3-iodo-

<b>Other names:</b>	3-Iodoheptane
<b>Inchi:</b>	InChI=1S/C7H15I/c1-3-5-6-7(8)4-2/h7H,3-6H2,1-2H3
<b>InchiKey:</b>	LVPGHYVXKUXNQR-UHFFFAOYSA-N
<b>Formula:</b>	C7H15I
<b>SMILES:</b>	CCCCCC(I)CC
<b>Mol. weight [g/mol]:</b>	226.10
<b>CAS:</b>	31294-92-5

## Physical Properties

Property code	Value	Unit	Source
gf	63.74	kJ/mol	Joback Method
hf	-116.22	kJ/mol	Joback Method
hfus	14.77	kJ/mol	Joback Method
hvap	40.16	kJ/mol	Joback Method
log10ws	-3.81		Crippen Method
logp	3.390		Crippen Method
mcvol	135.310	ml/mol	McGowan Method
pc	2758.46	kPa	Joback Method
rinpol	1131.00		NIST Webbook
rinpol	1071.00		NIST Webbook
ripol	1270.00		NIST Webbook
tb	452.26	K	Joback Method
tc	655.04	K	Joback Method
tf	211.71	K	Joback Method
vc	0.509	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	249.32	J/mol×K	452.26	Joback Method
cpg	306.24	J/mol×K	621.25	Joback Method
cpg	296.03	J/mol×K	587.45	Joback Method
cpg	285.26	J/mol×K	553.65	Joback Method
cpg	273.90	J/mol×K	519.85	Joback Method

cpg	261.93	J/mol×K	486.06	Joback Method
cpg	315.92	J/mol×K	655.04	Joback Method
dvisc	0.0003227	Paxs	452.26	Joback Method
dvisc	0.0004326	Paxs	412.17	Joback Method
dvisc	0.0006178	Paxs	372.08	Joback Method
dvisc	0.0009615	Paxs	331.99	Joback Method
dvisc	0.0016897	Paxs	291.89	Joback Method
dvisc	0.0035535	Paxs	251.80	Joback Method
dvisc	0.0099031	Paxs	211.71	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.47819e+01
Coeff. B	-4.02472e+03
Coeff. C	-7.10040e+01
Temperature range (K), min.	348.68
Temperature range (K), max.	495.98

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C31294925&Units=SI>

## Legend

**cpg:** Ideal gas heat capacity  
**dvisc:** Dynamic viscosity  
**gf:** Standard Gibbs free energy of formation  
**hf:** Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/41-293-1/Heptane-3-iodo.pdf>

Generated by Cheméo on 2024-04-25 18:22:58.780308478 +0000 UTC m=+16358627.700885793.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.