

# Isobutyl chloroformate

<b>Other names:</b>	Carbonochloridic acid, 2-methylpropyl ester Chlorocarbonic acid isobutyl ester Chloroformic acid isobutyl ester Formic acid, chloro-, isobutyl ester Isobutyl chlorocarbonate 2-Methylpropyl carbonochloridate 2-Methylpropyl chloroformate Isobutyl chloridocarbonate NSC 8429
<b>Inchi:</b>	InChI=1S/C5H9ClO2/c1-4(2)3-8-5(6)7/h4H,3H2,1-2H3
<b>InchiKey:</b>	YOETUEMZNOLGDB-UHFFFAOYSA-N
<b>Formula:</b>	C5H9ClO2
<b>SMILES:</b>	CC(C)COC(=O)Cl
<b>Mol. weight [g/mol]:</b>	136.58
<b>CAS:</b>	543-27-1

## Physical Properties

Property code	Value	Unit	Source
gf	-257.07	kJ/mol	Joback Method
hf	-412.35	kJ/mol	Joback Method
hfus	12.17	kJ/mol	Joback Method
hvap	39.88	kJ/mol	Joback Method
log10ws	-1.67		Crippen Method
logp	2.018		Crippen Method
mvol	100.990	ml/mol	McGowan Method
pc	3538.87	kPa	Joback Method
rinpol	808.00		NIST Webbook
tb	402.00	K	NIST Webbook
tc	617.43	K	Joback Method
tf	233.19	K	Joback Method
vc	0.383	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	186.92	J/mol×K	427.08	Joback Method
cpg	195.50	J/mol×K	458.81	Joback Method
cpg	203.77	J/mol×K	490.53	Joback Method
cpg	211.73	J/mol×K	522.26	Joback Method
cpg	219.39	J/mol×K	553.98	Joback Method
cpg	226.73	J/mol×K	585.71	Joback Method
cpg	233.77	J/mol×K	617.43	Joback Method
dvisc	0.0042591	Paxs	233.19	Joback Method
dvisc	0.0020958	Paxs	265.50	Joback Method
dvisc	0.0012028	Paxs	297.82	Joback Method
dvisc	0.0007696	Paxs	330.13	Joback Method
dvisc	0.0005332	Paxs	362.45	Joback Method
dvisc	0.0003923	Paxs	394.76	Joback Method
dvisc	0.0003024	Paxs	427.08	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C543271&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307i>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/41-311-0/Isobutyl-chloroformate.pdf>

Generated by Cheméo on 2024-04-20 16:10:16.126896763 +0000 UTC m=+15918665.047474107.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.