

# Dimethylmalonic acid, ethyl neopentyl ester

<b>Inchi:</b>	InChI=1S/C12H22O4/c1-7-15-9(13)12(5,6)10(14)16-8-11(2,3)4/h7-8H2,1-6H3
<b>InchiKey:</b>	YXUCGPFCFPXRAL-UHFFFAOYSA-N
<b>Formula:</b>	C12H22O4
<b>SMILES:</b>	CCOC(=O)C(C)(C)C(=O)OCC(C)(C)C
<b>Mol. weight [g/mol]:</b>	230.30

## Physical Properties

Property code	Value	Unit	Source
gf	-412.00	kJ/mol	Joback Method
hf	-798.11	kJ/mol	Joback Method
hfus	17.58	kJ/mol	Joback Method
hvap	58.03	kJ/mol	Joback Method
log10ws	-2.09		Crippen Method
logp	2.165		Crippen Method
mcvol	194.820	ml/mol	McGowan Method
pc	1992.98	kPa	Joback Method
rinpol	1270.00		NIST Webbook
tb	620.08	K	Joback Method
tc	815.26	K	Joback Method
tf	374.16	K	Joback Method
vc	0.734	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	521.98	J/molxK	620.08	Joback Method
cpg	537.75	J/molxK	652.61	Joback Method
cpg	552.63	J/molxK	685.14	Joback Method
cpg	566.64	J/molxK	717.67	Joback Method
cpg	579.82	J/molxK	750.20	Joback Method
cpg	592.18	J/molxK	782.73	Joback Method
cpg	603.78	J/molxK	815.26	Joback Method
dvisc	0.0019720	Paxs	374.16	Joback Method
dvisc	0.0009780	Paxs	415.15	Joback Method

dvisc	0.0005501	Paxs	456.13	Joback Method
dvisc	0.0003403	Paxs	497.12	Joback Method
dvisc	0.0002264	Paxs	538.11	Joback Method
dvisc	0.0001596	Paxs	579.09	Joback Method
dvisc	0.0001179	Paxs	620.08	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361743&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361743&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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