

# Succinic acid, cis-hex-3-enyl hexadecyl ester

**Inchi:** InChI=1S/C26H48O4/c1-3-5-7-9-10-11-12-13-14-15-16-17-18-20-24-30-26(28)22-21-25(29)  
**InchiKey:** VINSBSZIFVFZLW-VURMDHGXSA-N  
**Formula:** C26H48O4  
**SMILES:** CCC=CCCOC(=O)CCC(=O)OCCCCCCCCCCCCCCCCC  
**Mol. weight [g/mol]:** 424.66

## Physical Properties

Property code	Value	Unit	Source
gf	-219.58	kJ/mol	Joback Method
hf	-952.35	kJ/mol	Joback Method
hfus	68.87	kJ/mol	Joback Method
hvap	91.74	kJ/mol	Joback Method
log10ws	-8.28		Crippen Method
logp	7.691		Crippen Method
mcvol	387.780	ml/mol	McGowan Method
pc	784.63	kPa	Joback Method
rinsol	2926.00		NIST Webbook
tb	951.02	K	Joback Method
tc	1169.32	K	Joback Method
tf	522.02	K	Joback Method
vc	1.520	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1317.56	J/molxK	951.02	Joback Method
cpg	1406.89	J/molxK	1132.93	Joback Method
cpg	1391.79	J/molxK	1096.55	Joback Method
cpg	1375.37	J/molxK	1060.17	Joback Method
cpg	1357.57	J/molxK	1023.79	Joback Method
cpg	1338.32	J/molxK	987.40	Joback Method
cpg	1420.74	J/molxK	1169.32	Joback Method
dvisc	0.0000189	Paxs	951.02	Joback Method
dvisc	0.0000255	Paxs	879.52	Joback Method

dvisc	0.0000361	Paxs	808.02	Joback Method
dvisc	0.0000548	Paxs	736.52	Joback Method
dvisc	0.0000910	Paxs	665.02	Joback Method
dvisc	0.0001707	Paxs	593.52	Joback Method
dvisc	0.0003802	Paxs	522.02	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353418&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353418&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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