

Naphthalene, decahydro-2-methyl-

Other names:	Decahydro-2-methylnaphthalene 2-Methyldecalin Decahydronaphthalene, 2-methyl 2-Methyldecalin, isomer 4 2-Methyldecahydronaphthalene 2-Methyldecalin, isomer 1 2-Methyldecalin, isomer 2 2-Methyldecalin, isomer 3
Inchi:	InChI=1S/C11H20/c1-9-6-7-10-4-2-3-5-11(10)8-9/h9-11H,2-8H2,1H3
InchiKey:	GREARFRXIFVLGB-UHFFFAOYSA-N
Formula:	C11H20
SMILES:	CC1CCC2CCCCC2C1
Mol. weight [g/mol]:	152.28
CAS:	2958-76-1

Physical Properties

Property code	Value	Unit	Source
chl	-6829.36	kJ/mol	NIST Webbook
gf	107.13	kJ/mol	Joback Method
hf	-169.75	kJ/mol	Joback Method
hfus	13.19	kJ/mol	Joback Method
hvap	40.29	kJ/mol	Joback Method
log10ws	-3.49		Crippen Method
logp	3.613		Crippen Method
mcvol	144.130	ml/mol	McGowan Method
pc	2637.96	kPa	Joback Method
rinpol	1122.00		NIST Webbook
rinpol	1186.00		NIST Webbook
rinpol	1173.00		NIST Webbook
rinpol	1160.00		NIST Webbook
rinpol	1115.00		NIST Webbook
rinpol	1159.00		NIST Webbook
rinpol	1146.00		NIST Webbook
rinpol	1133.00		NIST Webbook
rinpol	1145.00		NIST Webbook
rinpol	1142.00		NIST Webbook
rinpol	1153.00		NIST Webbook

rinpol	1166.00		NIST Webbook
rinpol	1154.00		NIST Webbook
rinpol	1167.00		NIST Webbook
rinpol	1180.00		NIST Webbook
rinpol	1146.00		NIST Webbook
tb	474.00 ± 4.00	K	NIST Webbook
tb	477.00 ± 4.00	K	NIST Webbook
tc	695.39	K	Joback Method
tf	231.29	K	Joback Method
vc	0.532	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	333.53	J/mol×K	476.97	Joback Method
cpg	437.80	J/mol×K	658.98	Joback Method
cpg	419.46	J/mol×K	622.58	Joback Method
cpg	399.91	J/mol×K	586.18	Joback Method
cpg	379.11	J/mol×K	549.78	Joback Method
cpg	356.99	J/mol×K	513.37	Joback Method
cpg	454.99	J/mol×K	695.39	Joback Method
dvisc	0.0004102	Paxs	476.97	Joback Method
dvisc	0.0004917	Paxs	436.02	Joback Method
dvisc	0.0006120	Paxs	395.08	Joback Method
dvisc	0.0008011	Paxs	354.13	Joback Method
dvisc	0.0011253	Paxs	313.18	Joback Method
dvisc	0.0017507	Paxs	272.24	Joback Method
dvisc	0.0031852	Paxs	231.29	Joback Method

Sources

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2958761&Units=SI>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Legend

chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpolar:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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