

# Butane, 2-chloro-2-methyl-

<b>Other names:</b>	1,1-DIMETHYLPROPYL CHLORIDE 2-Chloro-2-methylbutane 2-Methyl-2-chlorobutane TERT-AMYL CHLORIDE TERT-PENTYL CHLORIDE Tertiary pentyl chloride
<b>Inchi:</b>	InChI=1S/C5H11Cl/c1-4-5(2,3)6/h4H2,1-3H3
<b>InchiKey:</b>	CRNIHJHMEQZAAS-UHFFFAOYSA-N
<b>Formula:</b>	C5H11Cl
<b>SMILES:</b>	CCC(C)(C)Cl
<b>Mol. weight [g/mol]:</b>	106.59
<b>CAS:</b>	594-36-5

## Physical Properties

Property code	Value	Unit	Source
chl	-3327.50 ± 8.40	kJ/mol	NIST Webbook
gf	-17.87	kJ/mol	Joback Method
hf	-171.02	kJ/mol	Joback Method
hfl	-235.00 ± 2.00	kJ/mol	NIST Webbook
hfus	5.49	kJ/mol	Joback Method
hvap	29.81	kJ/mol	Joback Method
log10ws	-2.51		Aqueous Solubility Prediction Method
log10ws	-2.51		Estimated Solubility Method
logp	2.414		Crippen Method
mcvol	93.550	ml/mol	McGowan Method
pc	3395.98	kPa	Joback Method
rinpol	657.00		NIST Webbook
rinpol	671.00		NIST Webbook
rinpol	657.00		NIST Webbook
rinpol	638.00		NIST Webbook
rinpol	642.00		NIST Webbook
rinpol	645.00		NIST Webbook
rinpol	650.00		NIST Webbook
rinpol	647.88		NIST Webbook
rinpol	657.00		NIST Webbook

rinpol	648.00			NIST Webbook
rinpol	664.00			NIST Webbook
rinpol	671.00			NIST Webbook
rinpol	638.00			NIST Webbook
rinpol	641.00			NIST Webbook
rinpol	647.89			NIST Webbook
ripol	826.00			NIST Webbook
ripol	837.00			NIST Webbook
ripol	849.00			NIST Webbook
tb	358.75 ± 0.50		K	NIST Webbook
tb	358.15 ± 2.00		K	NIST Webbook
tb	358.00 ± 2.00		K	NIST Webbook
tb	358.80		K	NIST Webbook
tb	358.80 ± 0.20		K	NIST Webbook
tc	532.51		K	Joback Method
tf	200.45 ± 0.50		K	NIST Webbook
tf	200.00 ± 5.00		K	NIST Webbook
tf	199.77		K	Aqueous Solubility Prediction Method
vc	0.353		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	199.01	J/molxK	501.76	Joback Method
cpg	190.56	J/molxK	471.01	Joback Method
cpg	181.62	J/molxK	440.25	Joback Method
cpg	172.18	J/molxK	409.50	Joback Method
cpg	162.20	J/molxK	378.75	Joback Method
cpg	151.66	J/molxK	348.00	Joback Method
cpg	207.00	J/molxK	532.51	Joback Method
dvisc	0.0090043	Paxs	178.45	Joback Method
dvisc	0.0003465	Paxs	348.00	Joback Method
dvisc	0.0004692	Paxs	319.74	Joback Method
dvisc	0.0006737	Paxs	291.48	Joback Method
dvisc	0.0010454	Paxs	263.23	Joback Method
dvisc	0.0018032	Paxs	234.97	Joback Method
dvisc	0.0036100	Paxs	206.71	Joback Method
hvapt	35.00	kJ/mol	338.00	NIST Webbook
hvapt	33.50 ± 0.84	kJ/mol	356.20	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54520e+01
Coeff. B	-3.44417e+03
Coeff. C	-4.08850e+01
Temperature range (K), min.	268.01
Temperature range (K), max.	380.53

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C*\ln(T) + D*T^2$
Coeff. A	8.20848e+01
Coeff. B	-6.87078e+03
Coeff. C	-1.00577e+01
Coeff. D	6.64130e-06
Temperature range (K), min.	280.15
Temperature range (K), max.	548.97

## Sources

<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C594365&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C594365&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>KDB:</b>	<a href="https://www.chemic.org/files/research/kdb/mol/mol1623.mol">https://www.chemic.org/files/research/kdb/mol/mol1623.mol</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.chemic.org/research/kdb/hcprop/showprop.php?cmpid=1623">https://www.chemic.org/research/kdb/hcprop/showprop.php?cmpid=1623</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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