

# Benzenebutanol, «alpha»-phenyl-

<b>Inchi:</b>	InChI=1S/C16H18O/c17-16(15-11-5-2-6-12-15)13-7-10-14-8-3-1-4-9-14/h1-6,8-9,11-12,1
<b>InchiKey:</b>	CAGIBUJGQRHYEW-UHFFFAOYSA-N
<b>Formula:</b>	C16H18O
<b>SMILES:</b>	OC(CCCc1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	226.31
<b>CAS:</b>	30078-89-8

## Physical Properties

Property code	Value	Unit	Source
gf	169.40	kJ/mol	Joback Method
hf	-58.02	kJ/mol	Joback Method
hfus	25.84	kJ/mol	Joback Method
hvap	72.05	kJ/mol	Joback Method
ie	8.50 ± 0.10	eV	NIST Webbook
log10ws	-4.45		Crippen Method
logp	3.743		Crippen Method
mcvol	194.650	ml/mol	McGowan Method
pc	2520.12	kPa	Joback Method
tb	710.58	K	Joback Method
tc	927.27	K	Joback Method
tf	368.74	K	Joback Method
vc	0.729	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	533.14	J/mol×K	710.58	Joback Method
cpg	548.08	J/mol×K	746.69	Joback Method
cpg	561.97	J/mol×K	782.81	Joback Method
cpg	574.86	J/mol×K	818.92	Joback Method
cpg	586.83	J/mol×K	855.04	Joback Method
cpg	597.94	J/mol×K	891.15	Joback Method
cpg	608.27	J/mol×K	927.27	Joback Method
dvisc	0.0038173	Paxs	368.74	Joback Method

dvisc	0.0010182	Paxs	425.71	Joback Method
dvisc	0.0003710	Paxs	482.69	Joback Method
dvisc	0.0001673	Paxs	539.66	Joback Method
dvisc	0.0000878	Paxs	596.63	Joback Method
dvisc	0.0000516	Paxs	653.61	Joback Method
dvisc	0.0000330	Paxs	710.58	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C30078898&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C30078898&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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