

# Ar-Turmerone

<b>Other names:</b>	Turmerone ar-tumerone Turmerone (-ar) Aromatic turmerone
<b>Inchi:</b>	InChI=1S/C15H20O/c1-11(2)9-15(16)10-13(4)14-7-5-12(3)6-8-14/h5-9,13H,10H2,1-4H3
<b>InchiKey:</b>	NAAJVHHFAXWBOK-UHFFFAOYSA-N
<b>Formula:</b>	C15H20O
<b>SMILES:</b>	<chem>CC(C)=CC(=O)CC(C)c1ccc(C)cc1</chem>
<b>Mol. weight [g/mol]:</b>	216.32
<b>CAS:</b>	532-65-0

## Physical Properties

Property code	Value	Unit	Source
gf	118.51	kJ/mol	Joback Method
hf	-138.30	kJ/mol	Joback Method
hfus	25.23	kJ/mol	Joback Method
hvap	58.32	kJ/mol	Joback Method
log10ws	-4.36		Crippen Method
logp	4.024		Crippen Method
mcvol	195.720	ml/mol	McGowan Method
pc	2040.07	kPa	Joback Method
rinpol	1659.00		NIST Webbook
rinpol	1664.00		NIST Webbook
rinpol	1664.00		NIST Webbook
rinpol	1665.00		NIST Webbook
rinpol	1638.00		NIST Webbook
rinpol	1672.30		NIST Webbook
rinpol	1669.00		NIST Webbook
rinpol	1636.00		NIST Webbook
rinpol	1640.00		NIST Webbook
rinpol	1631.00		NIST Webbook
rinpol	1632.00		NIST Webbook
rinpol	1661.00		NIST Webbook
rinpol	1657.00		NIST Webbook
rinpol	1658.00		NIST Webbook
rinpol	1658.00		NIST Webbook
rinpol	1637.00		NIST Webbook

rinpol	1662.00		NIST Webbook
rinpol	1667.00		NIST Webbook
rinpol	1616.00		NIST Webbook
rinpol	1669.00		NIST Webbook
rinpol	1664.00		NIST Webbook
rinpol	1628.00		NIST Webbook
rinpol	1638.00		NIST Webbook
rinpol	1670.00		NIST Webbook
tb	631.73	K	Joback Method
tc	847.25	K	Joback Method
tf	313.64	K	Joback Method
vc	0.749	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	497.60	J/mol×K	631.73	Joback Method
cpg	514.72	J/mol×K	667.65	Joback Method
cpg	530.78	J/mol×K	703.57	Joback Method
cpg	545.84	J/mol×K	739.49	Joback Method
cpg	559.95	J/mol×K	775.41	Joback Method
cpg	573.17	J/mol×K	811.33	Joback Method
cpg	585.56	J/mol×K	847.25	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C532650&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C532650&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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