

# Furan, 2,5-diethyltetrahydro-

<b>Other names:</b>	2,5-Diethyltetrahydrofuran 2,5-Diethyltetrahydrofurane
<b>Inchi:</b>	InChI=1S/C8H16O/c1-3-7-5-6-8(4-2)9-7/h7-8H,3-6H2,1-2H3
<b>InchiKey:</b>	YKWLEIXVUHRKEF-UHFFFAOYSA-N
<b>Formula:</b>	C8H16O
<b>SMILES:</b>	CCC1CCC(CC)O1
<b>Mol. weight [g/mol]:</b>	128.21
<b>CAS:</b>	41239-48-9

## Physical Properties

Property code	Value	Unit	Source
gf	-40.80	kJ/mol	Joback Method
hf	-300.31	kJ/mol	Joback Method
hfus	19.46	kJ/mol	Joback Method
hvap	37.86	kJ/mol	Joback Method
log10ws	-2.38		Crippen Method
logp	2.354		Crippen Method
mcvol	118.590	ml/mol	McGowan Method
pc	2928.17	kPa	Joback Method
rinpol	896.00		NIST Webbook
rinpol	896.00		NIST Webbook
rinpol	901.00		NIST Webbook
rinpol	896.00		NIST Webbook
rinpol	884.00		NIST Webbook
rinpol	884.00		NIST Webbook
rinpol	916.00		NIST Webbook
rinpol	902.00		NIST Webbook
rinpol	898.30		NIST Webbook
tb	420.00	K	Joback Method
tc	612.65	K	Joback Method
tf	213.15	K	Joback Method
vc	0.445	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	245.73	J/molxK	420.00	Joback Method
cpg	261.97	J/molxK	452.11	Joback Method
cpg	277.49	J/molxK	484.22	Joback Method
cpg	292.31	J/molxK	516.33	Joback Method
cpg	306.43	J/molxK	548.43	Joback Method
cpg	319.87	J/molxK	580.54	Joback Method
cpg	332.66	J/molxK	612.65	Joback Method
dvisc	0.0031621	Paxs	213.15	Joback Method
dvisc	0.0016960	Paxs	247.62	Joback Method
dvisc	0.0010592	Paxs	282.10	Joback Method
dvisc	0.0007330	Paxs	316.58	Joback Method
dvisc	0.0005452	Paxs	351.05	Joback Method
dvisc	0.0004276	Paxs	385.52	Joback Method
dvisc	0.0003490	Paxs	420.00	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C41239489&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C41239489&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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