

# 1,4-Benzenediol, monobenzoate

|                      |                                                                          |
|----------------------|--------------------------------------------------------------------------|
| Other names:         | 1,4-Diphenol, 1-benzoate<br>p-hydroxyphenyl benzoate                     |
| Inchi:               | InChI=1S/C13H10O3/c14-11-6-8-12(9-7-11)16-13(15)10-4-2-1-3-5-10/h1-9,14H |
| InchiKey:            | JFAXJRJMFOACBO-UHFFFAOYSA-N                                              |
| Formula:             | C13H10O3                                                                 |
| SMILES:              | O=C(Oc1ccc(O)cc1)c1ccccc1                                                |
| Mol. weight [g/mol]: | 214.22                                                                   |
| CAS:                 | 2444-19-1                                                                |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -105.14 | kJ/mol  | Joback Method  |
| hf            | -260.70 | kJ/mol  | Joback Method  |
| hfus          | 26.08   | kJ/mol  | Joback Method  |
| hvap          | 71.25   | kJ/mol  | Joback Method  |
| log10ws       | -3.11   |         | Crippen Method |
| logp          | 2.611   |         | Crippen Method |
| mcvol         | 159.820 | ml/mol  | McGowan Method |
| pc            | 3862.67 | kPa     | Joback Method  |
| tb            | 707.11  | K       | Joback Method  |
| tc            | 962.79  | K       | Joback Method  |
| tf            | 472.99  | K       | Joback Method  |
| vc            | 0.537   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 413.44 | J/mol×K | 707.11          | Joback Method |
| cpg           | 467.02 | J/mol×K | 920.18          | Joback Method |
| cpg           | 457.84 | J/mol×K | 877.56          | Joback Method |
| cpg           | 448.02 | J/mol×K | 834.95          | Joback Method |
| cpg           | 437.42 | J/mol×K | 792.34          | Joback Method |
| cpg           | 425.94 | J/mol×K | 749.72          | Joback Method |
| cpg           | 475.68 | J/mol×K | 962.79          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0000147 | Pa×s | 707.11 | Joback Method |
| dvisc | 0.0000213 | Pa×s | 668.09 | Joback Method |
| dvisc | 0.0000323 | Pa×s | 629.07 | Joback Method |
| dvisc | 0.0000517 | Pa×s | 590.05 | Joback Method |
| dvisc | 0.0000885 | Pa×s | 551.03 | Joback Method |
| dvisc | 0.0001645 | Pa×s | 512.01 | Joback Method |
| dvisc | 0.0003388 | Pa×s | 472.99 | Joback Method |

## Sources

|                        |                                                                                                                                             |
|------------------------|---------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2444191&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2444191&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                   |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                           |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                       |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                       |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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