

# Tricyclo[2.2.1.0<sup>2</sup>,<sup>6</sup>]-heptane, 3-bromo-

<b>Other names:</b>	Nortricyclyl bromide
<b>Inchi:</b>	InChI=1S/C7H9Br/c8-7-3-1-4-5(2-3)6(4)7/h3-7H,1-2H2
<b>InchiKey:</b>	RDIUINMODIQXGA-UHFFFAOYSA-N
<b>Formula:</b>	C7H9Br
<b>SMILES:</b>	BrC1C2CC3C(C2)C13
<b>Mol. weight [g/mol]:</b>	173.05
<b>CAS:</b>	695-02-3

## Physical Properties

Property code	Value	Unit	Source
gf	213.41	kJ/mol	Joback Method
hf	28.56	kJ/mol	Joback Method
hfus	19.92	kJ/mol	Joback Method
hvap	36.39	kJ/mol	Joback Method
log10ws	-2.01		Crippen Method
logp	2.036		Crippen Method
mcvol	94.410	ml/mol	McGowan Method
pc	4178.49	kPa	Joback Method
tb	428.06	K	Joback Method
tc	642.63	K	Joback Method
tf	280.83	K	Joback Method
vc	0.374	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	193.78	J/molxK	428.06	Joback Method
cpg	256.02	J/molxK	606.87	Joback Method
cpg	245.71	J/molxK	571.11	Joback Method
cpg	234.43	J/molxK	535.34	Joback Method
cpg	222.10	J/molxK	499.58	Joback Method
cpg	208.58	J/molxK	463.82	Joback Method
cpg	265.50	J/molxK	642.63	Joback Method
dvisc	0.0018989	Paxs	428.06	Joback Method

dvisc	0.0015978	Paxs	403.52	Joback Method
dvisc	0.0013147	Paxs	378.98	Joback Method
dvisc	0.0010530	Paxs	354.44	Joback Method
dvisc	0.0008159	Paxs	329.91	Joback Method
dvisc	0.0006069	Paxs	305.37	Joback Method
dvisc	0.0004286	Paxs	280.83	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	351.00 ± 1.00	K	2.70	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C695023&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C695023&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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