

# Urea, 1,3-diethyl-1-nitroso-

Inchi:	InChI=1S/C5H11N3O2/c1-3-6-5(9)8(4-2)7-10/h3-4H2,1-2H3,(H,6,9)
InchiKey:	DEWYOTJLJIGZHS-UHFFFAOYSA-N
Formula:	C5H11N3O2
SMILES:	CCNC(=O)N(CC)N=N
Mol. weight [g/mol]:	145.16
CAS:	49540-32-1

## Physical Properties

Property code	Value	Unit	Source
hf	-306.30	kJ/mol	Joback Method
hvap	51.05	kJ/mol	Joback Method
log10ws	-1.61		Crippen Method
logp	0.719		Crippen Method
mcvol	114.390	ml/mol	McGowan Method
pc	3708.97	kPa	Joback Method
tb	493.68	K	Joback Method
tc	675.01	K	Joback Method

## Sources

Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C49540321&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C49540321&amp;Units=SI</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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