

# 1-Bromo-2,4,6-triisopropylbenzene

<b>Inchi:</b>	InChI=1S/C15H23Br/c1-9(2)12-7-13(10(3)4)15(16)14(8-12)11(5)6/h7-11H,1-6H3
<b>InchiKey:</b>	FUMMYHVKFAHQST-UHFFFAOYSA-N
<b>Formula:</b>	C15H23Br
<b>SMILES:</b>	CC(C)c1cc(C(C)C)c(Br)c(C(C)C)c1
<b>Mol. weight [g/mol]:</b>	283.25
<b>CAS:</b>	21524-34-5

## Physical Properties

Property code	Value	Unit	Source
gf	165.94	kJ/mol	Joback Method
hf	-140.32	kJ/mol	Joback Method
hfus	22.20	kJ/mol	Joback Method
hvap	58.52	kJ/mol	Joback Method
log10ws	-6.19		Crippen Method
logp	5.819		Crippen Method
mcvol	215.950	ml/mol	McGowan Method
pc	1925.36	kPa	Joback Method
tb	649.06	K	Joback Method
tc	867.42	K	Joback Method
tf	337.59	K	Joback Method
vc	0.811	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	542.99	J/molxK	649.06	Joback Method
cpg	560.83	J/molxK	685.45	Joback Method
cpg	577.63	J/molxK	721.85	Joback Method
cpg	593.43	J/molxK	758.24	Joback Method
cpg	608.27	J/molxK	794.63	Joback Method
cpg	622.21	J/molxK	831.02	Joback Method
cpg	635.27	J/molxK	867.42	Joback Method
dvisc	0.0020784	Paxs	337.59	Joback Method
dvisc	0.0009434	Paxs	389.50	Joback Method

dvisc	0.0005156	Paxs	441.41	Joback Method
dvisc	0.0003200	Paxs	493.32	Joback Method
dvisc	0.0002175	Paxs	545.24	Joback Method
dvisc	0.0001581	Paxs	597.15	Joback Method
dvisc	0.0001209	Paxs	649.06	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C21524345&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C21524345&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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