

# Benzene, 1-ethyl-2,3-dimethyl-

<b>Other names:</b>	1,2-DIMETHYL-3-ETHYLBENZENE 1-Ethyl-2,3-dimethylbenzene 3-Ethyl-1,2-dimethylbenzene 3-Ethyl-o-xylene o-Xylene, 3-ethyl-
<b>Inchi:</b>	InChI=1S/C10H14/c1-4-10-7-5-6-8(2)9(10)3/h5-7H,4H2,1-3H3
<b>InchiKey:</b>	QUBBAXISAHIDNM-UHFFFAOYSA-N
<b>Formula:</b>	C10H14
<b>SMILES:</b>	CCc1cccc(C)c1C
<b>Mol. weight [g/mol]:</b>	134.22
<b>CAS:</b>	933-98-2

## Physical Properties

Property code	Value	Unit	Source
af	0.4070		KDB
chl	-5853.00 ± 2.60	kJ/mol	NIST Webbook
chl	-5855.40 ± 1.20	kJ/mol	NIST Webbook
gf	126.47	kJ/mol	Joback Method
hf	-36.14	kJ/mol	Joback Method
hfl	-83.00 ± 2.60	kJ/mol	NIST Webbook
hfl	-80.50 ± 1.30	kJ/mol	NIST Webbook
hfus	14.92	kJ/mol	Joback Method
hvap	54.90	kJ/mol	NIST Webbook
log10ws	-3.23		Crippen Method
logp	2.866		Crippen Method
mcvol	128.000	ml/mol	McGowan Method
pc	2929.00	kPa	KDB
rinpol	1094.00		NIST Webbook
rinpol	1072.00		NIST Webbook
rinpol	1095.00		NIST Webbook
rinpol	1090.00		NIST Webbook
rinpol	1096.50		NIST Webbook
rinpol	1093.00		NIST Webbook
rinpol	1084.00		NIST Webbook
rinpol	1107.00		NIST Webbook
rinpol	1104.00		NIST Webbook
rinpol	1088.00		NIST Webbook

rinpol	1085.00	NIST Webbook
rinpol	1113.10	NIST Webbook
rinpol	1088.00	NIST Webbook
rinpol	1087.00	NIST Webbook
rinpol	1095.40	NIST Webbook
rinpol	1094.00	NIST Webbook
rinpol	1101.00	NIST Webbook
rinpol	1120.00	NIST Webbook
rinpol	1094.00	NIST Webbook
rinpol	1095.00	NIST Webbook
rinpol	1083.50	NIST Webbook
rinpol	1084.30	NIST Webbook
rinpol	1080.77	NIST Webbook
rinpol	1104.00	NIST Webbook
rinpol	1106.00	NIST Webbook
rinpol	1109.00	NIST Webbook
rinpol	1108.00	NIST Webbook
rinpol	1110.00	NIST Webbook
rinpol	1113.00	NIST Webbook
rinpol	1115.00	NIST Webbook
rinpol	1119.00	NIST Webbook
rinpol	1051.00	NIST Webbook
rinpol	1088.10	NIST Webbook
rinpol	1087.00	NIST Webbook
rinpol	1125.00	NIST Webbook
rinpol	1093.15	NIST Webbook
rinpol	1094.00	NIST Webbook
rinpol	1096.50	NIST Webbook
rinpol	1094.00	NIST Webbook
rinpol	1085.80	NIST Webbook
rinpol	1088.40	NIST Webbook
rinpol	1090.00	NIST Webbook
rinpol	1087.00	NIST Webbook
rinpol	1094.00	NIST Webbook
rinpol	1082.00	NIST Webbook
rinpol	1089.00	NIST Webbook
rinpol	1061.00	NIST Webbook
rinpol	1081.00	NIST Webbook
rinpol	1090.00	NIST Webbook
rinpol	1099.00	NIST Webbook
rinpol	1095.00	NIST Webbook
rinpol	1106.00	NIST Webbook
rinpol	1106.00	NIST Webbook
rinpol	1093.00	NIST Webbook

ripol	1082.50		NIST Webbook
ripol	1072.00		NIST Webbook
ripol	1368.90		NIST Webbook
ripol	1395.00		NIST Webbook
ripol	1369.00		NIST Webbook
ripol	1394.80		NIST Webbook
ripol	1390.00		NIST Webbook
ripol	1369.00		NIST Webbook
ripol	1352.00		NIST Webbook
ripol	1362.00		NIST Webbook
ripol	1364.00		NIST Webbook
ripol	1369.00		NIST Webbook
ripol	1374.00		NIST Webbook
ripol	1362.00		NIST Webbook
ripol	1430.80		NIST Webbook
ripol	1394.80		NIST Webbook
ripol	1395.00		NIST Webbook
tb	462.90	K	KDB
tc	665.10	K	KDB
tf	223.61 ± 0.20	K	NIST Webbook
vc	0.446	m <sup>3</sup> /kmol	KDB
zc	0.2364390		KDB

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	322.46	J/molxK	637.55	Joback Method
cpg	310.90	J/molxK	603.01	Joback Method
cpg	298.71	J/molxK	568.47	Joback Method
cpg	285.86	J/molxK	533.92	Joback Method
cpg	272.33	J/molxK	499.38	Joback Method
cpg	258.11	J/molxK	464.84	Joback Method
cpg	333.41	J/molxK	672.09	Joback Method
dvisc	0.0017125	Paxs	253.92	Joback Method
dvisc	0.0002068	Paxs	464.84	Joback Method
dvisc	0.0002547	Paxs	429.69	Joback Method
dvisc	0.0003255	Paxs	394.53	Joback Method
dvisc	0.0004364	Paxs	359.38	Joback Method
dvisc	0.0006236	Paxs	324.23	Joback Method
dvisc	0.0009718	Paxs	289.07	Joback Method
hvapt	49.70	kJ/mol	420.50	NIST Webbook

rhoI

885.86

kg/m<sup>3</sup>

293.10

KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.43482e+01
Coeff. B	-3.94990e+03
Coeff. C	-6.11560e+01
Temperature range (K), min.	342.07
Temperature range (K), max.	498.25

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.31621e+01
Coeff. B	-9.06264e+03
Coeff. C	-9.75815e+00
Coeff. D	3.83702e-06
Temperature range (K), min.	223.64
Temperature range (K), max.	680.00

## Sources

**Crippen Method:**[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)**Joback Method:**[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)**KDB:**<https://www.thermo.com/files/research/kdb/mol/mol678.mol>**McGowan Method:**<http://link.springer.com/article/10.1007/BF02311772>**NIST Webbook:**<http://webbook.nist.gov/cgi/cbook.cgi?ID=C933982&Units=SI>**The Yaws Handbook of Vapor Pressure:**<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>**KDB Vapor Pressure Data:**<https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=678>**Crippen Method:**<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

# Legend

<b>af:</b>	Acentric Factor
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume
<b>zc:</b>	Critical Compressibility

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