

# Diethylmalonic acid, pentyl 3-phenylpropyl ester

Inchi:	InChI=1S/C21H32O4/c1-4-7-11-16-24-19(22)21(5-2,6-3)20(23)25-17-12-15-18-13-9-8-10
InchiKey:	QJILYJQZYQVKCC-UHFFFAOYSA-N
Formula:	C21H32O4
SMILES:	CCCCCOC(=O)C(CC)(CC)C(=O)OCCc1ccccc1
Mol. weight [g/mol]:	348.48

## Physical Properties

Property code	Value	Unit	Source
gf	-226.65	kJ/mol	Joback Method
hf	-738.59	kJ/mol	Joback Method
hfus	42.35	kJ/mol	Joback Method
hvap	81.63	kJ/mol	Joback Method
log10ws	-5.20		Crippen Method
logp	4.702		Crippen Method
mcvol	297.870	ml/mol	McGowan Method
pc	1282.83	kPa	Joback Method
rinsol	2310.00		NIST Webbook
tb	855.91	K	Joback Method
tc	1060.47	K	Joback Method
tf	499.59	K	Joback Method
vc	1.141	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	938.54	J/molxK	855.91	Joback Method
cpg	955.09	J/molxK	890.00	Joback Method
cpg	970.47	J/molxK	924.10	Joback Method
cpg	984.71	J/molxK	958.19	Joback Method
cpg	997.88	J/molxK	992.28	Joback Method
cpg	1010.03	J/molxK	1026.38	Joback Method
cpg	1021.19	J/molxK	1060.47	Joback Method
dvisc	0.0005772	Paxs	499.59	Joback Method
dvisc	0.0002862	Paxs	558.98	Joback Method

dvisc	0.0001624	Paxs	618.36	Joback Method
dvisc	0.0001018	Paxs	677.75	Joback Method
dvisc	0.0000688	Paxs	737.14	Joback Method
dvisc	0.0000493	Paxs	796.52	Joback Method
dvisc	0.0000370	Paxs	855.91	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369654&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369654&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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