

# 1-Octadecanamine, N-methyl-

<b>Other names:</b>	Methyl-n-octadecylamine Methyloctadecylamine Methylstearylamine N-Methyl-1-octadecanamine N-Methyl-N-stearylamine N-Methyl-n-octadecylamine N-Methyloctadecanamine N-Methyloctadecylamine N-Methylstearylamine N-Stearyl-N-methylamine NSC 66468 Octadecylamine, N-methyl- Octadecylmethylamine
<b>Inchi:</b>	InChI=1S/C19H41N/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-2/h20H,3-19H2
<b>InchiKey:</b>	SZEGKVHRCLBFKJ-UHFFFAOYSA-N
<b>Formula:</b>	C19H41N
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCNC
<b>Mol. weight [g/mol]:</b>	283.54
<b>CAS:</b>	2439-55-6

## Physical Properties

Property code	Value	Unit	Source
gf	198.49	kJ/mol	Joback Method
hf	-382.02	kJ/mol	Joback Method
hfus	50.07	kJ/mol	Joback Method
hvap	64.32	kJ/mol	Joback Method
log10ws	-6.97		Crippen Method
logp	6.467		Crippen Method
mcvol	288.550	ml/mol	McGowan Method
pc	1089.94	kPa	Joback Method
rinpol	2088.00		NIST Webbook
tb	684.29	K	Joback Method
tc	848.90	K	Joback Method
tf	314.65 ± 2.00	K	NIST Webbook
tf	314.65 ± 2.00	K	NIST Webbook
vc	1.135	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	850.95	J/mol×K	684.29	Joback Method
cpg	871.16	J/mol×K	711.72	Joback Method
cpg	890.51	J/mol×K	739.16	Joback Method
cpg	909.02	J/mol×K	766.59	Joback Method
cpg	926.72	J/mol×K	794.03	Joback Method
cpg	943.63	J/mol×K	821.46	Joback Method
cpg	959.79	J/mol×K	848.90	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.58073e+01
Coeff. B	-5.65798e+03
Coeff. C	-1.13552e+02
Temperature range (K), min.	478.12
Temperature range (K), max.	652.62

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2439556&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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