

Fumaric acid, monoamide, N-(5-chloro-2-methoxyphenyl)-, 2-pentyl ester

Inchi:	InChI=1S/C16H20ClNO4/c1-4-5-11(2)22-16(20)9-8-15(19)18-13-10-12(17)6-7-14(13)21-
InchiKey:	XWDWFFOEUQCGF-CMDGGGOBGSA-N
Formula:	C16H20ClNO4
SMILES:	CCCC(C)OC(=O)C=CC(=O)Nc1cc(Cl)ccc1OC
Mol. weight [g/mol]:	325.79

Physical Properties

Property code	Value	Unit	Source
gf	-135.61	kJ/mol	Joback Method
hf	-499.91	kJ/mol	Joback Method
hfus	42.01	kJ/mol	Joback Method
hvap	83.51	kJ/mol	Joback Method
log10ws	-4.24		Crippen Method
logp	3.575		Crippen Method
mvol	245.340	ml/mol	McGowan Method
pc	1849.92	kPa	Joback Method
rinpol	2718.00		NIST Webbook
tb	846.02	K	Joback Method
tc	1062.40	K	Joback Method
tf	528.36	K	Joback Method
vc	0.929	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	702.56	J/mol×K	846.02	Joback Method
cpg	715.39	J/mol×K	882.08	Joback Method
cpg	727.19	J/mol×K	918.15	Joback Method
cpg	738.00	J/mol×K	954.21	Joback Method
cpg	747.84	J/mol×K	990.27	Joback Method
cpg	756.74	J/mol×K	1026.34	Joback Method
cpg	764.72	J/mol×K	1062.40	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U357498&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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