

# Butanoyl chloride, 4-chloro-

<b>Other names:</b>	«gamma»-Chlorobutyryl chloride 4-Chlorobutyryl chloride 4-Chlorobutyl chloride «gamma»-Chloro-n-butyryl chloride 4-Chlorobutyroyl chloride
<b>Inchi:</b>	InChI=1S/C4H6Cl2O/c5-3-1-2-4(6)7/h1-3H2
<b>InchiKey:</b>	CDIIZULDSLKBKV-UHFFFAOYSA-N
<b>Formula:</b>	C4H6Cl2O
<b>SMILES:</b>	O=C(Cl)CCCCl
<b>Mol. weight [g/mol]:</b>	141.00
<b>CAS:</b>	4635-59-0

## Physical Properties

Property code	Value	Unit	Source
gf	-169.98	kJ/mol	Joback Method
hf	-269.95	kJ/mol	Joback Method
h <sub>fus</sub>	16.11	kJ/mol	Joback Method
h <sub>vap</sub>	40.01	kJ/mol	Joback Method
log <sub>10</sub> w <sub>s</sub>	-1.58		Crippen Method
log <sub>p</sub>	1.771		Crippen Method
m <sub>cvol</sub>	93.270	ml/mol	McGowan Method
pc	3843.54	kPa	Joback Method
r <sub>inpol</sub>	959.70		NIST Webbook
tb	446.70	K	NIST Webbook
tc	615.08	K	Joback Method
tf	244.61	K	Joback Method
vc	0.363	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
c <sub>pg</sub>	152.74	J/mol×K	419.65	Joback Method
c <sub>pg</sub>	159.49	J/mol×K	452.22	Joback Method
c <sub>pg</sub>	165.90	J/mol×K	484.79	Joback Method

cpg	172.00	J/molxK	517.37	Joback Method
cpg	177.79	J/molxK	549.94	Joback Method
cpg	183.28	J/molxK	582.51	Joback Method
cpg	188.48	J/molxK	615.08	Joback Method
dvisc	0.0036532	Paxs	244.61	Joback Method
dvisc	0.0020955	Paxs	273.78	Joback Method
dvisc	0.0013378	Paxs	302.96	Joback Method
dvisc	0.0009241	Paxs	332.13	Joback Method
dvisc	0.0006777	Paxs	361.30	Joback Method
dvisc	0.0005205	Paxs	390.48	Joback Method
dvisc	0.0004147	Paxs	419.65	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4635590&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4635590&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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