

# 1-Propanone, 2,2-dimethyl-1-(4-phenoxyphenyl)-

<b>Other names:</b>	4'-Phenoxypropalophenone tert-Butyl 4-phenoxyphenyl ketone
<b>Inchi:</b>	InChI=1S/C17H18O2/c1-17(2,3)16(18)13-9-11-15(12-10-13)19-14-7-5-4-6-8-14/h4-12H,1
<b>InchiKey:</b>	ITEMAFJPIQQQMO-UHFFFAOYSA-N
<b>Formula:</b>	C17H18O2
<b>SMILES:</b>	CC(C)(C)C(=O)c1ccc(Oc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	254.32
<b>CAS:</b>	55814-54-5

## Physical Properties

Property code	Value	Unit	Source
gf	76.37	kJ/mol	Joback Method
hf	-186.17	kJ/mol	Joback Method
hfus	22.85	kJ/mol	Joback Method
hvap	66.51	kJ/mol	Joback Method
log10ws	-4.78		Crippen Method
logp	4.708		Crippen Method
mvol	210.310	ml/mol	McGowan Method
pc	2187.68	kPa	Joback Method
tb	719.76	K	Joback Method
tc	962.04	K	Joback Method
tf	421.29	K	Joback Method
vc	0.784	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	573.82	J/molxK	719.76	Joback Method
cpg	645.14	J/molxK	921.66	Joback Method
cpg	633.29	J/molxK	881.28	Joback Method
cpg	620.32	J/molxK	840.90	Joback Method
cpg	606.15	J/molxK	800.52	Joback Method
cpg	590.68	J/molxK	760.14	Joback Method
cpg	655.96	J/molxK	962.04	Joback Method

dvisc	0.0000906	Paxs	719.76	Joback Method
dvisc	0.0001177	Paxs	670.01	Joback Method
dvisc	0.0001595	Paxs	620.27	Joback Method
dvisc	0.0002278	Paxs	570.52	Joback Method
dvisc	0.0003482	Paxs	520.78	Joback Method
dvisc	0.0005824	Paxs	471.04	Joback Method
dvisc	0.0010996	Paxs	421.29	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	448.20	K	0.40	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55814545&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55814545&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/42-118-4/1-Propanone-2-2-dimethyl-1-4-phenoxyphenyl.pdf>

Generated by Cheméo on 2024-04-28 00:05:40.835890578 +0000 UTC m=+16551989.756467899.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.