

# Benzene, 1,3-dibromo-5-(bromomethyl)-

<b>Other names:</b>	3,5-Dibromobenzyl bromide
<b>Inchi:</b>	InChI=1S/C7H5Br3/c8-4-5-1-6(9)3-7(10)2-5/h1-3H,4H2
<b>InchiKey:</b>	PWTFRUXTAFBWBW-UHFFFAOYSA-N
<b>Formula:</b>	C7H5Br3
<b>SMILES:</b>	BrCc1cc(Br)cc(Br)c1
<b>Mol. weight [g/mol]:</b>	328.83
<b>CAS:</b>	56908-88-4

## Physical Properties

Property code	Value	Unit	Source
gf	144.17	kJ/mol	Joback Method
hf	104.77	kJ/mol	Joback Method
hfus	23.00	kJ/mol	Joback Method
hvap	54.08	kJ/mol	Joback Method
log10ws	-5.11		Crippen Method
logp	4.106		Crippen Method
mcvol	138.230	ml/mol	McGowan Method
pc	5235.81	kPa	Joback Method
tb	594.68	K	Joback Method
tc	861.95	K	Joback Method
tf	399.51	K	Joback Method
vc	0.505	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	229.92	J/molxK	594.68	Joback Method
cpg	237.90	J/molxK	639.22	Joback Method
cpg	245.16	J/molxK	683.77	Joback Method
cpg	251.79	J/molxK	728.31	Joback Method
cpg	257.86	J/molxK	772.86	Joback Method
cpg	263.45	J/molxK	817.40	Joback Method
cpg	268.64	J/molxK	861.95	Joback Method
dvisc	0.0012345	Paxs	399.51	Joback Method

dvisc	0.0008869	Paxs	432.04	Joback Method
dvisc	0.0006674	Paxs	464.57	Joback Method
dvisc	0.0005212	Paxs	497.09	Joback Method
dvisc	0.0004196	Paxs	529.62	Joback Method
dvisc	0.0003464	Paxs	562.15	Joback Method
dvisc	0.0002920	Paxs	594.68	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C56908884&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C56908884&amp;Units=SI</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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