

# Fumaric acid, monoamide, N-(2-fluorophenyl)-, 3,5-difluorophenyl ester

<b>Other names:</b>	Fumaric acid, monoamide, N-(2-fluorophenyl)-, 3,5-fluorophenyl ester
<b>Inchi:</b>	InChI=1S/C16H10F3NO3/c17-10-7-11(18)9-12(8-10)23-16(22)6-5-15(21)20-14-4-2-1-3-1
<b>InchiKey:</b>	ZAPGXAMCOLOTES-AATRIKPKSA-N
<b>Formula:</b>	C16H10F3NO3
<b>SMILES:</b>	O=C(C=CC(=O)Oc1cc(F)cc(F)c1)Nc1ccccc1F
<b>Mol. weight [g/mol]:</b>	321.25

## Physical Properties

Property code	Value	Unit	Source
gf	-497.89	kJ/mol	Joback Method
hf	-709.94	kJ/mol	Joback Method
hfus	43.04	kJ/mol	Joback Method
hvap	77.59	kJ/mol	Joback Method
log10ws	-4.49		Crippen Method
logp	3.204		Crippen Method
mvol	208.780	ml/mol	McGowan Method
pc	2271.90	kPa	Joback Method
rinpol	2402.00		NIST Webbook
rinpol	2402.00		NIST Webbook
tb	816.08	K	Joback Method
tc	1036.75	K	Joback Method
tf	531.92	K	Joback Method
vc	0.815	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	577.70	J/molxK	816.08	Joback Method
cpg	588.43	J/molxK	852.86	Joback Method
cpg	598.24	J/molxK	889.64	Joback Method
cpg	607.19	J/molxK	926.41	Joback Method
cpg	615.32	J/molxK	963.19	Joback Method
cpg	622.66	J/molxK	999.97	Joback Method
cpg	629.29	J/molxK	1036.75	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357406&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357406&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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