

# 1-Nitro-2-nitrosobenzene

**Inchi:** InChI=1S/C6H4N2O3/c9-7-5-3-1-2-4-6(5)8(10)11/h1-4H  
**InchiKey:** NSGVPXKWKUZSAY-UHFFFAOYSA-N  
**Formula:** C6H4N2O3  
**SMILES:** O=Nc1ccccc1[N+](=O)[O-]  
**Mol. weight [g/mol]:** 152.11  
**CAS:** 612-29-3

## Physical Properties

Property code	Value	Unit	Source
chs	-3038.70 ± 1.90	kJ/mol	NIST Webbook
hf	-121.06	kJ/mol	Joback Method
hfs	102.30	kJ/mol	NIST Webbook
hsub	54.00	kJ/mol	NIST Webbook
hvap	57.58	kJ/mol	Joback Method
log10ws	-2.88		Crippen Method
logp	1.993		Crippen Method
mcvol	100.610	ml/mol	McGowan Method
pc	4540.80	kPa	Joback Method
tb	583.58	K	Joback Method
tc	830.33	K	Joback Method

## Sources

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307l>  
**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)  
**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)  
**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C612293&Units=SI>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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