

Acetanilide, 2',4'-diamino

Other names:	N-(2,4-diaminophenyl)acetamide
Inchi:	InChI=1S/C8H11N3O/c1-5(12)11-8-3-2-6(9)4-7(8)10/h2-4H,9-10H2,1H3,(H,11,12)
InchiKey:	XLXOPVQPBPLVHP-UHFFFAOYSA-N
Formula:	C8H11N3O
SMILES:	CC(=O)Nc1ccc(N)cc1N
Mol. weight [g/mol]:	165.19
CAS:	6373-15-5

Physical Properties

Property code	Value	Unit	Source
gf	203.00	kJ/mol	Joback Method
hf	13.61	kJ/mol	Joback Method
hfus	26.83	kJ/mol	Joback Method
hvap	71.47	kJ/mol	Joback Method
log10ws	-0.94		Crippen Method
logp	0.809		Crippen Method
mcvol	131.330	ml/mol	McGowan Method
pc	4474.22	kPa	Joback Method
tb	668.18	K	Joback Method
tc	908.23	K	Joback Method
tf	500.49	K	Joback Method
vc	0.474	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	337.47	J/mol×K	668.18	Joback Method
cpg	348.30	J/mol×K	708.19	Joback Method
cpg	358.34	J/mol×K	748.20	Joback Method
cpg	367.62	J/mol×K	788.20	Joback Method
cpg	376.16	J/mol×K	828.21	Joback Method
cpg	384.01	J/mol×K	868.22	Joback Method
cpg	391.19	J/mol×K	908.23	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C6373155&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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