

# Dimethylmalonic acid, di(2-naphthylmethyl) ester

Inchi:	InChI=1S/C27H24O4/c1-27(2,25(28)30-17-19-11-13-21-7-3-5-9-23(21)15-19)26(29)31-1
InchiKey:	FTMPIEMKSULARP-UHFFFAOYSA-N
Formula:	C27H24O4
SMILES:	CC(C)(C(=O)OCc1ccc2ccccc2c1)C(=O)OCc1ccc2ccccc2c1
Mol. weight [g/mol]:	412.48

## Physical Properties

Property code	Value	Unit	Source
gf	130.32	kJ/mol	Joback Method
hf	-266.70	kJ/mol	Joback Method
hfus	45.19	kJ/mol	Joback Method
hvap	101.87	kJ/mol	Joback Method
log10ws	-8.06		Crippen Method
logp	5.806		Crippen Method
mcvol	319.730	ml/mol	McGowan Method
pc	1505.81	kPa	Joback Method
rinpol	3506.00		NIST Webbook
tb	1067.79	K	Joback Method
tc	1323.35	K	Joback Method
tf	684.07	K	Joback Method
vc	1.212	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1027.42	J/molxK	1067.79	Joback Method
cpg	1091.10	J/molxK	1280.75	Joback Method
cpg	1078.80	J/molxK	1238.16	Joback Method
cpg	1066.48	J/molxK	1195.57	Joback Method
cpg	1053.94	J/molxK	1152.98	Joback Method
cpg	1040.98	J/molxK	1110.38	Joback Method
cpg	1103.56	J/molxK	1323.35	Joback Method
dvisc	0.0000719	Paxs	1067.79	Joback Method
dvisc	0.0000870	Paxs	1003.84	Joback Method

dvisc	0.0001080	Paxs	939.88	Joback Method
dvisc	0.0001385	Paxs	875.93	Joback Method
dvisc	0.0001847	Paxs	811.98	Joback Method
dvisc	0.0002586	Paxs	748.02	Joback Method
dvisc	0.0003858	Paxs	684.07	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U363865&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U363865&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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