

# Diethylmalonic acid, 2-naphthyl pentyl ester

<b>Inchi:</b>	InChI=1S/C22H28O4/c1-4-7-10-15-25-20(23)22(5-2,6-3)21(24)26-19-14-13-17-11-8-9-12
<b>InchiKey:</b>	SYJVKEKSZSDAPP-UHFFFAOYSA-N
<b>Formula:</b>	C22H28O4
<b>SMILES:</b>	CCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccc2ccccc2c1
<b>Mol. weight [g/mol]:</b>	356.46

## Physical Properties

Property code	Value	Unit	Source
gf	-121.21	kJ/mol	Joback Method
hf	-579.63	kJ/mol	Joback Method
hfus	41.57	kJ/mol	Joback Method
hvap	86.16	kJ/mol	Joback Method
log10ws	-6.40		Crippen Method
logp	5.285		Crippen Method
mcvol	292.500	ml/mol	McGowan Method
pc	1415.44	kPa	Joback Method
rinpol	2588.00		NIST Webbook
tb	902.75	K	Joback Method
tc	1121.90	K	Joback Method
tf	556.08	K	Joback Method
vc	1.119	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	920.61	J/molxK	902.75	Joback Method
cpg	935.72	J/molxK	939.27	Joback Method
cpg	949.75	J/molxK	975.80	Joback Method
cpg	962.77	J/molxK	1012.32	Joback Method
cpg	974.87	J/molxK	1048.85	Joback Method
cpg	986.12	J/molxK	1085.37	Joback Method
cpg	996.61	J/molxK	1121.90	Joback Method
dvisc	0.0005280	Paxs	556.08	Joback Method
dvisc	0.0003154	Paxs	613.86	Joback Method

dvisc	0.0002059	Paxs	671.64	Joback Method
dvisc	0.0001438	Paxs	729.41	Joback Method
dvisc	0.0001059	Paxs	787.19	Joback Method
dvisc	0.0000813	Paxs	844.97	Joback Method
dvisc	0.0000645	Paxs	902.75	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369883&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369883&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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