

# Urea, 1-(2-fluoroethyl)-1-nitroso-3-(tetrahydro-2h-thiopy s,s-dioxide

Inchi:  
InchiKey:

InChI=1S/C8H14FN3O4S/c9-3-4-12(11-14)8(13)10-7-1-5-17(15,16)6-2-7/h7H,1-6H2,(H,11,13,15,16)2H,2-4H,5-6H,7-8H,10-12H,14H,15-16H;S(=O)(=O)c1ccccc1

LKCXGNLVAZPIG-UHFFFAOYSA-N

Formula: C8H14FN3O4S  
SMILES: O=NN(CCF)C(=O)NC1CCS(=O)(=O)CC1  
Mol. weight [g/mol]: 267.28  
CAS: 32319-89-4

## Physical Properties

Property code	Value	Unit	Source
hf	-959.97	kJ/mol	Joback Method
hvap	74.97	kJ/mol	Joback Method
log10ws	-1.56		Crippen Method
logp	0.226		Crippen Method
mcvol	175.660	ml/mol	McGowan Method
pc	3727.11	kPa	Joback Method
tb	607.97	K	Joback Method
tc	794.30	K	Joback Method

## Sources

- NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C32319894&Units=SI>
- Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci9903071>
- Crippen Method: [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)
- Joback Method: [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)
- McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

## Legend

- hf: Enthalpy of formation at standard conditions
- hvap: Enthalpy of vaporization at standard conditions
- log10ws: Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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